One-Pot Approach to N‑Quinolyl 3′/4′-Biaryl Carboxamides by Microwave-Assisted Suzuki−Miyaura Coupling and N‑Boc Deprotection

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S Supporting Information

[AB](#page-9-0)STRACT: N[-Quinolyl b](#page-9-0)iaryl carboxamides have received tremendous attention for their notable biological properties. Here we have described a general protocol for the preparation of N-quinolyl 3′/4′-biaryl carboxamides by microwave-assisted Suzuki−Miyaura cross-coupling reaction and N-Boc deprotection in one pot. This method, which did not require acids, was used to produce a series of N-quinolyl 3′/4′-biaryl carboxamides with excellent functional group tolerance and high yields (70% to 95%).

ENTRODUCTION

The biaryl carboxamides play important roles as vital building blocks in the synthesis of a diversity of drugs such as anticancer,^{1-4} anti-inflammatory, 5,6 anxiolytic, 7 Alzheimer's, 8 and anemia therapeutic agents.⁹ Among these biaryl carbox-amides, N[-qui](#page-9-0)nol[y](#page-10-0)l (Q) 3'/4'-bi[ar](#page-9-0)yl carboxam[id](#page-10-0)[e](#page-10-0) is a unique substructure in medicinal chemis[tr](#page-10-0)y that has shown the potential to antagonize bacteria, 10 promote weight $loss, 11$ promote differentiation of adult human cardiac progenitor cells, 12 and activate TRPV1 (transie[nt](#page-10-0) receptor potential vani[llo](#page-10-0)id type 1) ion channel.^{13,14}

Although a variety of procedures are available for preparing N-Q 2′-biaryl [carbo](#page-10-0)xamides, methods available for the synthesis of N-Q 3′/4′-biaryl carboxamides are still limited. N-Q 3′/4′-biaryl carboxamides are often synthesized starting from halogensubstituted aromatic esters and aromatic boronic acids via three steps including Suzuki−Miyaura coupling, ester hydrolysis, and amide formation (Scheme 1, b).^{13–15} However, these multistep processes are usually time-consuming and tedious procedures. Recently, Jana and Singh have [report](#page-10-0)ed an elegant example of preparing 2′,4′-difluoro-N-(quinolin-8-yl)[1,1′-biphenyl]-4-carboxamide via the direct Suzuki−Miyaura cross-coupling reaction of 4-bromo-N-(quinolin-8-yl)benzamide and (2,4 difluorophenyl)boronic acid.¹⁶ However, this method requires a large amount of aromatic boronic acids (2.0 equiv). More importantly, the amide bear[ing](#page-10-0) a 8-amioquinolinyl moiety can strongly chelate the palladium catalyst, which may poison the catalyst and result in the high loading of catalyst. Thus, it is still desirable to develop low cost protocols for the synthesis of N-Q Scheme 1. Selected Methods To Synthesize N-Q 3′/4′-Biaryl Carboxamides

3′/4′-biaryl carboxamides. Moreover, during the course of our study on preparing N-(pyridin-2-ylmethyl)biphenyl-4-sulfonamides, 17 we found that the N-Boc protection material could be easily prepared in higher yields. Further, the one-pot method is less ti[me](#page-10-0)-consuming and gives a higher yield than the stepwise approach. Hence, the one-pot method has drawn the attention of chemists.¹⁸ Herein, we report a one-pot, microwave-assisted Suzuki−Miyaura cross-coupling and N-Boc-deprotection ap-

Received: July 19, 2016 Published: September 20, 2016 Table 1. Optimization of Pd(PPh₃)₄-Catalyzed Suzuki–Miyaura Coupling and Water-Promoted N-Boc-Deprotecting One-Pot Reaction^{a, $\frac{a}{b}$}

^aReactions conditions: (1) 3a (0.2 mmol), 4a (0.2 mmol), catalyst was Pd(PPh₃)₄, base (0.4 mmol), solvent 4 mL (3 mL of organic solvent and 1 mL of H₂O); (2) under N₂. ^bIsolated yield. ^cCatalyst was Pd/C (10%) ^dConventional heating.

proach to prepare a variety of N-Q 3′/4′-biaryl carboxamides with excellent functional group tolerance and high yields.

Although the palladium-catalyzed Suzuki−Miyaura reactions were found to be the best approaches to synthesize biaryl carboxamides,¹⁹ the direct Suzuki−Miyaura coupling to N-8 quinolyl 3′/4′-biaryl carboxamides took more time and wasted 1.0 equiv of [bor](#page-10-0)onic acid. Thus, it is obvious that the catalyst could be inactivated by the N-8-quinolyl to some extent.^{15,19e,20} Thus, it is beneficial to prevent the N-8-quinolyl group from chelating the catalyst in order to facilitate the ec[onomic](#page-10-0) conversion of starting materials. Even more important, in our continuous efforts on the application of microwave irradiation in the target-oriented synthesis,²¹ our present aim is to develop a one-pot strategy toward N-Q 3′/4′-biaryl carboxamides via sequential $Pd(PPh_3)_4$ -cataly[zed](#page-10-0) Suzuki–Miyaura coupling and water-promoted N-Boc deprotection under microwave irradiation.

RESULTS AND DISCUSSION

Although the conditions of coupling and the deprotection of amide N-Boc are known separately, 22 it is still a big challenge to find the optimum conditions to perform Suzuki−Miyaura coupling and amide N-Boc deprot[ect](#page-10-0)ion sequentially in a onepot manner under microwave irradiation. For our initial studies, the reaction of bromide 3a with 1.0 equiv of phenylboronic aicd 4a in the presence of $Pd(PPh₃)₄$ and NaOAc under microwave irradiation was chosen as a benchmark reaction (Table 1; for the optimization of amide N-Boc deprotection see Table S1). To our delight, 78% of coupling and deprotecting product (5aa) was

obtained at a temperature of 80 °C in 6 min (Table 1, entry 1). Process monitoring indicated that 3a could not be completely converted into the Suzuki−Miyaura coupling product 5b, but the deprotection of 5b and the remaining 3a was smooth. Therefore, in order to improve the yield of 5aa, the reaction time was prolonged to promote the first-step Suzuki−Miyaura coupling. Thus, the desired product 5aa was obtained in the same yield of 94% in 8 and 10 min (entries 2 and 3). However, the yield of 5aa was reduced to 84% with a temperature of 85 $^{\circ}$ C (entry 4). In addition, different solvents were screened, such as water mixed with dioxane, DME, DMF, 2-Me-THF, and 2-propanol (entries 2 and 5−8). The mixture of water and dioxane ($v/v = 1:3$) was shown to produce the best results. Subsequently, the effect of base on this reaction was explored, and among these bases (NaOAc, Na₂CO₃, K₃PO₄, and Cs₂CO₃), NaOAc was identified as the best one (entries 2 and 9−11). In addition, we noted that in the absence of base (entry 12) or catalyst (entry 15) the coupling reaction would not take place, but the deprotection was not affected. Finally, the effect of palladium loading was invested. We found that decreasing the loading of catalyst to 1.5 mol % (entry 13) and 1.0 mol % (entry 14) gave an 87% and 81% yield of 5aa, respectively, while, a small amount of 5aa was obtained when 2.0 and 10 mol % of Pd/C was used (entries 16 and 17). Compared with conventional heating (entry 18), microwave irradiation could significantly accelerate the reaction and notably improve the yield of the product. As a result, the combination of 2.0 mol % of $Pd(PPh_3)_4$, 1.0 equiv of phenylboronic acid, and 2.0 equiv of NaOAc in dioxane mixed water with a temperature of 80 °C for 8 min for the first step followed by increasing the reaction

Table 2. Scope of 3a–c and (Hetero)arylboronic Acid $(4)^{a,b}$

^aReactions conditions:3a−c (0.2 mmol), 4 (0.2 mmol), Pd(PPh₃)₄ (2 mol %), NaOAc (0.4 mmol), dioxane/H₂O = 3 mL:1 mL, T₂ = 120 °C, t₂ = 8 min, protected by N_2 . b Isolated yield.

temperature to 120 °C over 8 min for the second step were chosen as the optimal conditions.

Once the optimized reaction conditions were identified, the scope and limitations of this one-pot process were explored. A variety of aromatic boronic acids and bromine-substituted N-Q arylamides were applied. Various substituted (hetero) arylboronic acids were examined, the results are shown in Table 2. The reactions between 3a and (hetero)arylboronic acids (4) always proceeded smoothly. Both electron-withdrawing and electron-donating groups, such as methoxycarbonyl (5am),

trifluoromethyl (5ak) and methyl (5ab), tert-butyl (5ac), methoxy (5ad), halogen (5ae−ah), and trifluoromethoxy (5aj) afforded the desired products in good to excellent yields. More importantly, the reaction was proven to be well tolerant of valuable but unstable groups, such as hydroxyl (5ai) and acetyl (5al). In addition, disubstituted and heteroarylboronic acids were also investigated and afforded the corresponding products (5an−as). To increase the scope of our one-pot reaction, the 3b and (hetero)arylboronic acids (4) were tested. Very similar results were obtained in the presence of the meta substituent N-Q benzamide (3b). The reaction of phenylboronic aid gave 5ba in 95% yields. The reaction of electron-rich phenylboronic acids with 4-tert-butyl and 4-methoxy substituents led to 5bb and 5bc in 95% and 94% yield, respectively. The halogen-substituted phenylboronic acids (5bd−be) could also give excellent yields. Even the electron-deficient phenylboronic acids, with methoxycarbonyl and trifluoromethyl groups, were found to be suitable partners for the one-pot reaction and gave 5bf and 5bg in good yields. Of note, the 2,4-dichloro-substituted phenylboronic and 4-pyridinylboronic acids could be successfully converted to the products 5bh−bi. Next, several reactions using 3c as the starting materials were performed. The target products 5ca−cd were obtained in 84−94% yields using 2.0 mol % of $Pd(PPh₃)₄$ as the catalyst under microwave irradiation. Furthermore, the structure of 5ba was confirmed by X-ray data, and it showed that the phenyl group was at the meta position (Figure 1).

Figure 1. ORTEP diagrams of 5ba (the thermal ellipsoids are drawn at the 50% probability level).

Notably, the one-pot reaction was not limited to the use of N-Q five-membered heteroaryl carboxamides. The reaction of (hetero)arylboronic acids (4) with 3d−f (containing furanyl, thiophenyl, and thiazolyl, respectively) also produced the desired products 6 in good to excellent yield (Table 3). It was obvious that 3d could produce more products than only 3e and 3f.

To further examine the one-pot reaction and rapidly expand our unique compound collection, we also carried out the reaction between 3g and (hetero)arylboronic acids (4). Accordingly, the target-oriented products, biaryl carboxamide pyrabactin (Py) analogues as abscisic acid (ABA) agonists (7aa−ag), were obtained in 86−95% yield (Table 4). The analogue 7aa has been found to be active on the ABA receptor PYR1 via a subsequent molecular simulation stud[y \(Figur](#page-4-0)e 2). The pyridine ring can form conservative $\pi-\pi$ stacking interactions with residue Y126 (Figure 2B). The amide gro[up can no](#page-4-0)t only form a hydrogen bond with residue E100 but also form another hydrogen bond [with resid](#page-4-0)ue K65 to stabilize the binding mode. However, the biphenyl group makes 7aa bind with P94 much more closely than bromine atom. As shown in Table 5, the estimated binding free energy of 7aa is −4.42 kcal/mol, which is nearly equal to that of ABA (−4.34 kcal/mol) but h[igher tha](#page-4-0)n that of pyrabactin (−7.12 kcal/mol).

^aReactions conditions:3**d−f** (0.2 mmol), 4 (0.2 mmol), Pd(PPh₃)₄ (2 mol %), NaOAc (0.4 mmol), dioxane/H₂O = 3 mL:1 mL, T_2 = 120 $^{\circ}C$, $t_2 = 8$ min, protected by N₂. ^bIsolated yield.

The molecular mechanics Poisson−Boltzmann surface area (MM-PBSA) method in the AMBER12 package was employed to perform the free energy analyses. The overall objective of the MM-PBSA method is to calculate the free energy difference between two states that most often represent the bound and unbound states of two solvated molecules. The free energy of the ligand binding with the receptor, ΔG_{bind} , is calculated from the difference between the free energy of the receptor−ligand complex (G_{complex}) and the sum of the free energies of the unbound receptor (G_{receptor}) and ligand (G_{liquid}) as shown in eq 1.

$$
\Delta G_{\text{bind}} = G_{\text{complex}} - (G_{\text{ligand}} + G_{\text{receptor}}) \tag{1}
$$

The binding free energy ΔG_bind includes three items: MM gasphase binding energy (ΔE_{MM}), solvation free energy (ΔG_{sol}), and entropy contribution $(-T\Delta S)$. The sum of molecular mechanical gas-phase binding energy (ΔE_{MM}) and solvation free energy (ΔG_{sol}) is denoted by the binding energy (ΔE_{bind}). The $\Delta G_{\rm bind}$ was estimated from $\Delta E_{\rm bind}$ and $-T\Delta S$ in eq 2. The $\Delta E_{\rm bind}$ is calculated from ΔE_{MM} and ΔG_{sol} in eq 3.

$$
\Delta G_{\text{bind}} = \Delta E_{\text{bind}} - T\Delta S \tag{2}
$$

$$
\Delta E_{\text{bind}} = \Delta E_{\text{MM}} + \Delta G_{\text{sol}} \tag{3}
$$

The ΔE_{MM} is calculated by eq 4 where ΔE_{ele} and ΔE_{VDW} represent the electrostatic and van der Waals interactions. The solvation free energy ΔG_{sol} consists of two parts: the electrostatic contribution to the solvation free energy (ΔG_{PB}) and nonelectrostatic contribution to the solvation free energy (ΔG_{np}) as described in eq 5.

$$
\Delta E_{\text{MM}} = \Delta E_{\text{ele}} + \Delta E_{\text{VDW}} \tag{4}
$$

$$
\Delta G_{\text{sol}} = \Delta G_{\text{PB}} + \Delta G_{\text{np}} \tag{5}
$$

$$
\Delta S = \Delta S_{\text{conf}} + \Delta S_{\text{sol}} \tag{6}
$$

Table 4. Scope of 3g and (Hetero)arylboronic Acid $(4)^{a,b}$

^aReactions conditions:3g (0.2 mmol), 4 (0.2 mmol), Pd(PPh₃)₄ (2 mol %), NaOAc (0.4 mmol), dioxane/H₂O = 3 mL:1 mL, T₂ = 120 °C, t₂ = 8 min, protected by N_2 . b Isolated yield.

Figure 2. Computational modeling of Py (A) and 7aa (B) in PYR1 (PDB code: 3QN1).

The ΔS_{sol} and ΔS_{conf} are the solvation entropy and the conformational entropy change in eq 6. In the binding process, the conformational entropy change is related to the change of the number of rotatable bonds during [the b](#page-3-0)inding process, and the solvation entropy is related to the tendency of water molecules to minimize their contacts with hydrophobic groups in protein. When an empirical solvation model is developed, one calculates the parameters against the available experimental data without accounting for the detailed structural information, which means to average the overall solvation contributions from both solvent and solute. Therefore, the computational procedure that was used to evaluate the entropic contribution $(-T\Delta S)$ to the binding free energy was the same as in our previous publication.²³

$$
-T\Delta S_{\rm conf} = w(\Delta N_{\rm rot})\tag{7}
$$

The contribution to the binding free energy from the conformational entropy change is proportional to the number (ΔN_{rot}) of the lost rotatable bonds during the binding in which w is the scaling factor (eq 7). This adjustable parameter (w) was calibrated to be 1 kcal/mol for the PYR1 proteins. We note that the w value of 1 kcal/mol used in the present study is the same as that used previously by other researchers.²⁴ The adjustment of the w value usually does not change the qualitative order of the calculated binding free energies calculated for a series of compounds binding with a given type of protein.²⁵

■ CONCLUSION

In summary, a one-pot microwave-assisted Suzuki−Miyaura cross-coupling reaction and N-Boc deprotection for preparing a variety of N-Q 3′/4′-biaryl carboxamides was developed. A series of N-Q biaryl carboxamides was obtained in good to excellent yields. In addition, the broad substrate scopes and excellent reactivity make the strategy operationally concise and facilitate rapid library construction of potential pyrabactin analogues as abscisic acid analogues.

EXPERIMENTAL SECTION

General Information. All reagents and solvents were obtained commercially except when otherwise noted. Thin-layer chromatography (TLC) analysis was used to monitor the reaction, which was carried out on silica plates. Flash column chromatography was performed using silica gel (200–300 mesh). ¹H spectra were recorded in CDCl₃ or DMSO- d_6 on 400 or 600 MHz NMR spectrometers and resonances (δ) are given in parts per million relative to tetramethylsilane. Data are reported as follows: chemical shift, multiplicity ($s =$ singlet, $d =$ doublet, t $=$ triplet, $m =$ multiplet, $q =$ quarternary), coupling constants (Hz), and integration. ¹³C spectra were recorded in CDCl₃ or DMSO- d_6 on 100 or 150 MHz NMR spectrometers and resonances (δ) are given in ppm. High-resolution mass spectra (HRMS) were analyzed by a TOF analyzer. Microwave irradiation reactions were carried out on a Smith synthesizer instrument (the temperature of reaction system was controlled by the wall infrared sensor, while the pressure was regulated by a noninvasive pressure sensor). All products reported showed ¹H and 13 C NMR spectra in agreement with the assigned structures.

General Procedure for the Synthesis of Compounds 3a−g. The synthesis of 3aa−ag is representative: compound 2 (amino, 10 mmol) and DMAP (3 mmol) were placed in a 100 mL, two-necked reaction flask, and the flask was flushed with nitrogen. Dichloromethane

Table 5. Binding Free Energies (kcal/m[ol](#page-10-0)) Calculated for the PYR1 with ABA, Py, and 7aa

(40 mL), triethylamine (12 mmol), and 1 (acid, 11 mmol) were added, and the mixture was stirred at room temperature for 18 h. The resulting mixture was then quenched with water. The mixture was extracted with dichloromethane, and the combined organic layer was dried over sodium sulfate. Concentration in vacuum followed by silica gel column purification (petroleum ether/acetone = 20:2) gave 3 aa-ag.²⁶⁻²⁸

The synthesis of 3a−g is representative: Boc anhydride (6 mmol) was added to a solution of 3aa−ag (3 mmol) and DMAP (4.5 [mmo](#page-10-0)l) in $CH₂Cl₂$ (40 mL), and the reaction mixture was stirred overnight. The reaction mixture was quenched with satd aq $NH₄Cl$ (40 mL) and extracted with CH_2Cl_2 (3 × 40 mL). The combined organic extracts were dried (Na_2SO_4) , concentrated under reduced pressure, and purified by column chromatography (petroleum ether/acetone = 20:2, $R_f = 0.15$) to give 3a–g.^{29–31}

Typical $Pd(PPh_3)_4$ -Catalyzed and Water-Promoted One-Pot Strategy to N-Q Bia[ryl Ca](#page-10-0)rboxamides 5−7. Compound 3 (0.2 mmol) and arylboronic acids (4, 0.2 mmol) were dissolved in dioxane (3 mL) and $H₂O$ (1 mL) in a microwave tube under a nitrogen atmosphere. $Pd(PPh₃)₄$ (2 mol %, 4.6 mg) and sodium acetate (0.4 mmol) were added, and the reaction mixture was irradiated in a microwave apparatus at 80 °C for 8−30 min. Then the temperature was increased to 120 °C for another 8 min. After the reaction mixture was cooled to ambient temperature, the product was concentrated, and the crude mixture was purified by column chromatography on silica gel (petroleum ether/ a cetone = 20:1.5) to the desired product.

4-Bromo-1-(quinolin-8-yl)benzamide (3aa): white solid (petroleum ether/acetone = 20:1, R_f = 0.15); 3.043g, yield 93%; mp 128-129 $^{\circ}$ C;^{32 1}H NMR (600 MHz, DMSO) δ 10.67 (s, 1H), 8.99 (dd, J = 4.2, 1.2 Hz, 1H), 8.71 (d, J = 7.8 Hz, 1H), 8.47 (dd, J = 8.4, 1.2 Hz, 1H), 7.99 $(d, J = 8.4 \text{ Hz}, 2H), 7.84 (d, J = 8.4 \text{ Hz}, 2H), 7.77 (d, J = 8.2 \text{ Hz}, 1H),$ $(d, J = 8.4 \text{ Hz}, 2H), 7.84 (d, J = 8.4 \text{ Hz}, 2H), 7.77 (d, J = 8.2 \text{ Hz}, 1H),$ $(d, J = 8.4 \text{ Hz}, 2H), 7.84 (d, J = 8.4 \text{ Hz}, 2H), 7.77 (d, J = 8.2 \text{ Hz}, 1H),$ 7.71−7.65 (m, 2H); 13C NMR (150 MHz, DMSO) δ 163.5, 149.1, 138.2, 136.6, 133.8, 133.4, 131.9, 129.1, 127.7, 126.9, 125.9, 122.4, 122.2, 116.7; HRMS (ESI) m/z [M + H]⁺ calcd for C₁₆H₁₂BrN₂O 327.0133, found 327.0121.

3-Bromo-1-(quinolin-8-yl)benzamide (3ab): white solid (petroleum ether/acetone = 20:1, R_f = 0.15); 3.01 g, yield 92%; mp 99–100 $\rm{^{\circ}C_3^{33}}$ ¹H NMR (600 MHz, DMSO) δ 10.71 (s, 1H), 9.04 (s, 1H), 8.72 $(d, J = 7.8 \text{ Hz}, 1\text{ H}), 8.52 (d, J = 8.4 \text{ Hz}, 1\text{ H}), 8.24 (s, 1\text{ H}), 8.09 (d, J = 7.2$ Hz[, 1H](#page-10-0)), 7.93 (d, J = 7.8 Hz, 1H), 7.83 (d, J = 7.8 Hz, 1H), 7.76−7.70 $(m, 2H)$, 7.65 $(t, J = 7.8 \text{ Hz}, 1H)$; ¹³C NMR (150 MHz, DMSO) δ 163.0, 149.1, 138.4, 136.7, 136.6, 134.7, 133.7, 131.0, 130.0, 127.8, 126.9, 125.9, 122.7, 122.3, 122.2, 117.2; HRMS (ESI) m/z [M + H]⁺ calcd for $C_{16}H_{12}BrN_2O$ 327.0133, found 327.0123.

4-bromo-1-(quinolin-8-yl)-1-naphthamide (3ac): white solid (petroleum ether/acetone = 20:1, R_f = 0.15); 3.395 g, yield 90%; mp 198− 199 °C; ¹H NMR (600 MHz, CDCl₃) δ 10.41 (s, 1H), 9.04 (s, 1H), 8.76 $(s, 1H)$, 8.54 $(s, 1H)$, 8.36 $(d, J = 7.8 \text{ Hz}, 1H)$, 8.23 $(s, 1H)$, 7.91 $(s, 1H)$, 7.78 (s, 1H), 7.65 (m, 4H), 7.49 (s, 1H); ¹³C NMR (150 MHz, CDCl₃) δ 166.9, 148.3, 138.3, 136.4, 134.4, 134.3, 132.1, 131.2, 129.0, 128.0, 127.8, 127.4, 127.3, 126.0, 125.9, 125.5, 122.2, 121.7, 116.9; HRMS (ESI) m/z [M + H]⁺ calcd for C₂₀H₁₄BrN₂O 377.0290, found 377.0278.

5-Bromo-1-(quinolin-8-yl)furan-2-carboxamide (3ad): white solid (petroleum ether/acetone = 20:1, R_f = 0.15); 2.601 g, yield 82%; mp 109−110 °C;^{34 1}H NMR (600 MHz, DMSO) δ 10.54 (s, 1H), 9.02 (dd, $J = 4.2, 1.2$ Hz, 1H), 8.68 (d, $J = 7.8$ Hz, 1H), 8.48 (dd, $J = 8.4, 1.2$ Hz, 1H), 7.76 (d, [J](#page-10-0) = 8.4 Hz, 1H), 7.70 (dd, J = 8.4, 4.2 Hz, 1H), 7.65 (t, J = 7.8 Hz, 1H), 7.43 (d, J = 3.6 Hz, 1H), 6.94 (d, J = 3.6 Hz, 1H); ¹³C NMR (150 MHz, DMSO) δ 154.3, 149.3, 149.1, 137.8, 136.8, 133.3, 127.8, 127.0, 125.6, 122.5, 122.4, 118.0, 116.4, 115.1; HRMS (ESI) m/z [M + $[H]^+$ calcd for $C_{14}H_{10}BrN_2O_2$ 316.9926, found 316.9915.

5-Bromo-N-(quinolin-8-yl)thiophene-2-carboxamide (3ae): white solid (petroleum ether/acetone = 20:1, R_f = 0.15); 2.699 g, yield 81%; mp 156–157 °C;^{35 1}H NMR (600 MHz, DMSO) δ 10.58 (s, 1H), 9.00 $(d, J = 4.2 \text{ Hz}, 1H), 8.50 (d, J = 7.8 \text{ Hz}, 2H), 7.90 (d, J = 4.2 \text{ Hz}, 1H),$ 7.80 (d, J = 8.4 H[z, 1](#page-10-0)H), 7.70 (dd, J = 8.4, 4.2 Hz, 1H), 7.66 (t, J = 7.8 Hz, 1H), 7.43 (d, J = 4.2 Hz, 1H); ¹³C NMR (150 MHz, CDCl₃) δ 159.0, 147.7, 141.2, 137.6, 137.4, 133.4, 130.9, 128.8, 128.1, 127.7, 122.2, 121.6, 119.0, 118.0; HRMS (ESI) m/z [M + H]⁺ calcd for C₁₄H₁₀BrN₂OS 332.9697, found 332.9665.

5-Bromo-N-(quinolin-8-yl)thiazole-2-carboxamide (3af): white solid (petroleum ether/acetone = 20:3, R_f = 0.15); 2.707 g, yield 81%; mp 137−138 °C; ¹H NMR (600 MHz, CĎCl₃) *δ* 10.55 (s, 1H), 8.87 (d, J = 4.2. Hz, 1H), 8.79−8.72 (m, 1H), 8.22 (d, J = 8.4 Hz, 2H), 7.61−7.56 $(m, 2H)$, 7.52 (dd, J = 8.4, 4.2 Hz, 1H); ¹³C NMR (150 MHz, CDCl₃) δ 157.2, 156.0, 148.5, 141.3, 138.1, 137.9, 136.4, 133.4, 127.8, 127.2, 122.4, 121.9, 116.7; HRMS (ESI) m/z $[M + H]^+$ calcd for $C_{13}H_9BrN_3OS$ 333.9650, found 333.9641.

4-Bromo-N-(pyridin-2-ylmethyl)benzamide (3ag): white solid (petroleum ether/acetone = 20:1, $R_f = 0.15$); 2.737 g, yield 94%; mp 112−113 °C;³⁶ ¹H NMR (600 MHz, CDCl₃) δ 8.57 (d, J = 4.2 Hz, 1H), 7.76 (d, J = 8.4 Hz, 2H), 7.72 (m, 2H), 7.59 (d, J = 8.4 Hz, 2H), 7.35 (d, J $= 7.8$ Hz, 1H[\),](#page-10-0) 7.25 (s, 1H), 4.76 (d, J = 4.8 Hz, 2H); ¹³C NMR (150 MHz, DMSO) δ 165.5, 158.6, 148.9, 136.7, 133.3, 131.4, 129.4, 125.1, 122.1, 121.0, 44.8; HRMS (ESI) m/z [M + H]⁺ calcd for C₁₃H₁₂BrN₂O 291.0133, found 291.0146.

tert-Butyl (4-bromobenzoyl)(quinolin-8-yl)carbamate (3a): white solid (petroleum ether/acetone = 20:1, R_f = 0.15); 1.180 g, yield 92%; mp 159−160 °C; ¹H NMR (600 MHz, DMSO) δ 8.90 (d, J = 4.2 Hz, 1H), 8.47 (d, $J = 8.4$ Hz, 1H), 8.06 (d, $J = 8.4$ Hz, 1H), 7.82 (d, $J = 7.2$ Hz, 1H), 7.76 (s, 4H), 7.68 (t, J = 7.8 Hz, 1H), 7.61 (dd, J = 8.4, 4.2 Hz, 1H), 1.14 (s, 9H); 13C NMR (150 MHz, DMSO) δ 171.7, 152.4, 150.7, 143.3, 136.5, 136.2, 131.3, 130.0, 129.3, 128.6, 128.5, 126.4, 125.0, 122.0, 82.7, 27.0; HRMS (ESI) m/z [M + H]⁺ calcd for C₂₁H₂₀BrN₂O₃ 427.0657, found 427.0630.

tert-Butyl (3-bromobenzoyl)(quinolin-8-yl)carbamate(3b): white solid (petroleum ether/acetone = 20:1, R_f = 0.15); 1.154 g, yield 90%; mp 125−126 °C; ¹H NMR (600 MHz, DMSO) δ 8.91 (d, J = 3.0 Hz, 1H), 8.48 (d, J = 7.8 Hz, 1H), 8.07 (d, J = 7.8 Hz, 1H), 8.00 (s, 1H), 7.86 $(d, J = 7.2 \text{ Hz}, 1H), 7.80 \text{ (m, 2H)}, 7.69 \text{ (t, } J = 7.8 \text{ Hz}, 1H), 7.61 \text{ (dd, } J =$ 8.4, 4.2 Hz, 1H), 7.51 (t, J = 7.8 Hz, 1H), 1.14 (s, 9H); ¹³C NMR (150 MHz, DMSO) δ 171.1, 152.3, 150.8, 143.3, 139.3, 136.5, 136.3, 134.0, 130.6, 130.4, 129.5, 128.7, 128.6, 126.8, 126.4, 122.1, 121.3, 82.9, 27.0; HRMS (ESI) m/z [M + H]⁺ calcd for C₂₁H₂₀BrN₂O₃ 427.0657, found 427.0635.

tert-Butyl (4-bromo-1-naphthoyl)(quinolin-8-yl)carbamate (3c): white solid (petroleum ether/acetone = 20:1, R_f = 0.15); 1.289 g, yield 90%; mp 159−160 °C; ¹H NMR (600 MHz, ČDCl₃) δ 9.00 (s, 1H), 8.65 (s, 1H), 8.33 (s, 1H), 8.24 (d, J = 7.8 Hz, 1H), 7.95 (s, 1H), 7.91 (d, J = 7.8 Hz, 1H), 7.86 (d, J = 7.8 Hz, 1H), 7.80 (d, J = 7.2 Hz, 1H), 7.65 $(d, J = 6.6 \text{ Hz}, 4\text{H}), 7.48 \text{ (s, 1H)}, 0.90 \text{ (s, 9H)};$ ¹³C NMR (150 MHz, CDCl3) δ 171.8, 152.6, 150.5, 136.3, 131.8, 131.6, 129.3, 129.2, 128.9, 128.6, 127.8, 127.7, 127.4, 126.3, 126.2, 125.4, 125.0, 121.7, 83.5, 27.1; HRMS (ESI) m/z [M + H]⁺ calcd for $C_{25}H_{22}BrN_2O_3$ 477.0814, found 477.0800.

tert-Butyl (5-bromofuran-2-carbonyl) (quinolin-8-yl)carbamate (3d): white solid (petroleum ether/acetone = 20:1, R_f = 0.15); 1.089 g, yield 87%; mp 1 $\overline{11-112}$ °C; $^1\overline{\rm H}$ NMR (600 MHz, DMSO) δ 8.88 (s, 1H), 8.47 (d, J = 8.4 Hz, 1H), 8.05 (d, J = 8.4 Hz, 1H), 7.75 (d, J = 7.2 Hz, 1H), 7.66 (t, J = 7.8 Hz, 1H), 7.60 (dd, J = 8.4, 4.2 Hz, 1H), 7.25 (d, J $= 3.0$ Hz, 1H), 6.90 (d, J = 3.0 Hz, 1H), 1.32 (s, 9H); ¹³C NMR (150 MHz, DMSO) δ 162.4, 155.0, 151.9, 150.9, 145.8, 143.2, 136.5, 135.8, 135.7, 129.4, 129.1, 128.6, 126.4, 122.2, 83.4, 27.2; HRMS (ESI) m/z $[M + H]^{+}$ calcd for $C_{19}H_{18}BrN_2O_4$ 417.0450, found 417.0461.

tert-Butyl (5-bromothiophene-2-carbonyl)(quinolin-8-yl) *carbamate (3e):* white solid (petroleum ether/acetone = 20:1, R_f = 0.15); 1.118 g, yield 86%; mp 92−93 °C; ¹ H NMR (600 MHz, DMSO) δ 8.91 (d, J = 4.2 Hz, 1H), 8.47 (d, J = 8.4 Hz, 1H), 8.05 (d, J = 8.4 Hz, 1H), 7.77 (d, J = 7.2 Hz, 1H), 7.66 (t, J = 7.8 Hz, 1H), 7.61 (dd, J = 8.4, 4.2 Hz, 1H), 7.39 (d, J = 4.2 Hz, 1H), 7.28 (d, J = 4.2 Hz, 1H), 1.26 (s, 9H); ¹³C NMR (150 MHz, CDCl₃) δ 164.4, 153.0, 150.4, 143.9, 140.4, 136.8, 136.1, 133.1, 130.0, 129.0, 128.9, 128.4, 126.2, 121.7, 119.9, 83.4, 27.6; HRMS (ESI) m/z [M + H]⁺ calcd for C₁₉H₁₈BrN₂O₃S 433.0222, found 433.0211.

tert-Butyl (5-bromothiazole-2-carbonyl)(quinolin-8-yl)carbamate (3f): white solid (petroleum ether/acetone =20:3, $R_f = 0.15$); 1.081 g, yield 83%; mp 111−112 °C; ¹H NMR (600 MHz, DMSO) δ 8.92 (d, J= 4.2 Hz, 1H), 8.49 (d, J = 8.4 Hz, 1H), 8.15 (s, 1H), 8.09 (d, J = 8.4 Hz, 1H), 7.86 (d, J = 7.2 Hz, 1H), 7.69 (t, J = 7.8 Hz, 1H), 7.63 (dd, J = 8.4, 4.2 Hz, 1H), 1.26 (s, 9H); 13C NMR (150 MHz, DMSO) δ 162.3, 155.0,

151.9, 150.9, 145.8, 143.2, 136.5, 135.8, 135.7, 129.4, 129.1, 128.6, 126.4, 122.2, 83.3, 27.2; HRMS (ESI) m/z [M + H]⁺ calcd for C₁₈H₁₇BrN₃O₃S 434.0174, found 434.0135.

tert-Butyl (4-bromobenzoyl)(pyridin-2-ylmethyl)carbamate (3g): white solid (petroleum ether/acetone = 20:1, R_f = 0.15); 1.080 g, yield 92%; mp 82−83 °C; ¹ H NMR (600 MHz, DMSO) δ 8.50 (d, J = 4.8 Hz, 1H), 7.79 (t, J = 7.8 Hz, 1H), 7.71 (d, J = 8.4 Hz, 2H), 7.56 (d, J = 8.4 Hz, 2H), 7.35 (d, J = 7.8 Hz, 1H), 7.30−7.25 (m, 1H), 5.01 (s, 2H), 1.10 (s, 9H); ¹³C NMR (150 MHz, CDCl₃) δ 172.2, 156.8, 153.0, 149.2, 136.4, 136.3, 131.1, 129.3, 125.4, 122.0, 120.8, 83.3, 50.1, 27.4; HRMS (ESI) m/z [M + H]⁺ calcd for C₁₈H₂₀BrN₂O₃ 391.0657, found 391.0659.

tert-Butyl [1,1′-biphenyl]-4-carbonyl(quinolin-8-yl)carbamate (5*a*): white solid (petroleum ether/acetone = 20:1, R_f = 0.15); 75.6 mg, yield 89%; mp 155−156 °C; ¹H NMR (600 MHz, DMSO) *δ* 8.93 $(d, J = 4.2 \text{ Hz}, 1\text{H})$, 8.48 $(d, J = 8.4 \text{ Hz}, 1\text{H})$, 8.06 $(d, J = 8.4 \text{ Hz}, 1\text{H})$, 7.91 (d, J = 7.2 Hz, 2H), 7.85 (d, J = 7.2 Hz, 2H), 7.82 (d, J = 7.2 Hz, 1H), 7.77 (d, J = 7.2 Hz, 2H), 7.69 (t, J = 7.8 Hz, 1H), 7.65−7.59 (m, 1H), 7.52 (t, J = 7.2 Hz, 2H), 7.44 (t, J = 7.2 Hz, 1H), 1.15 (s, 9H); ¹³C NMR (150 MHz, DMSO) δ 172.3, 152.7, 150.7, 143.5, 143.1, 139.1, 136.8, 136.4, 135.8, 129.3, 129.1, 128.8, 128.7, 128.4, 128.2, 126.9, 126.4, 122.0, 82.5, 27.0; HRMS (ESI) m/z [M + H]⁺ calcd for C₂₇H₂₅N₂O₃ 425.1865, found 425.1867.

N-(Quinolin-8-yl)[1,1'-biphenyl]-4-carboxamide (5aa): white solid (petroleum ether/acetone = 20:1, R_f = 0.15); 61.0 mg, yield 94%; mp 153–154 °C;³⁷ ¹H NMR (600 MHz, DMSO) δ 10.74 (s, 1H), 9.02 (s, 1H), 8.77 (d, J = 7.8 Hz, 1H), 8.49 (d, J = 8.4 Hz, 1H), 8.14 (d, J = 7.2 Hz, 2H), 7.9[4 \(](#page-10-0)d, J = 7.2 Hz, 2H), 7.78 (m, 3H), 7.74−7.65 (m, 2H), 7.54 (t, J = 7.2 Hz, 2H), 7.46 (t, J = 7.2 Hz, 1H); ¹³C NMR (150 MHz, DMSO) δ 164.1, 149.2, 143.6, 138.9, 138.3, 136.8, 134.0, 133.1, 129.1, 128.3, 127.9, 127.7, 127.2, 127.1, 127.0, 122.4, 116.6; HRMS (ESI) m/z $[M + H]^{+}$ calcd for $C_{22}H_{17}N_{2}O$ 325.1341, found 325.1335.

4′-Methyl-1-(quinolin-8-yl)[1,1′-biphenyl]-4-carboxamide (5ab): white solid (petroleum ether/acetone = 20:1, R_f = 0.15); 63.6 mg, yield 94%; mp 154−155 °C; ¹H NMR (600 MHz, CDCl₃) (δ ppm): 10.73 (s, 1H), 9.01 (s, 1H), 8.77 (d, J = 7.8 Hz, 1H), 8.49 (d, J = 8.4 Hz, 1H), 8.12 (d, J = 7.8 Hz, 2H), 7.91 (d, J = 7.8 Hz, 2H), 7.77 (d, J = 8.4 Hz, 1H), 7.68 (t, J = 8.4 Hz, 4H), 7.34 (d, J = 7.8 Hz, 2H), 2.38 (s, 3H); ¹³C NMR (150 MHz, DMSO) δ 164.1, 149.2, 143.5, 138.2, 137.8, 136.8, 136.0, 134.0, 132.8, 129.7, 127.8, 127.7, 127.1, 126.9, 126.8, 122.4, 122.3, 116.5, 20.7; HRMS (ESI) m/z [M + H]⁺ calcd for C₂₃H₁₉N₂O 339.1497, found 339.1488.

4′-tert-Butyl-1-(quinolin-8-yl)[1,1′-biphenyl]-4-carboxamide (**5ac**): white solid (petroleum ether/acetone = 20:1, R_f = 0.15); 72.3 mg, yield 95%; mp 148−149 °C; ¹H NMR (600 MHz, DMSO) δ 10.73 (s, 1H), 9.01 (s, 1H), 8.77 (d, J = 7.2 Hz, 1H), 8.49 (d, J = 7.8 Hz, 1H), 8.12 $(d, J = 7.2 \text{ Hz}, 2H), 7.91 (d, J = 7.2 \text{ Hz}, 2H), 7.76 (d, J = 7.8 \text{ Hz}, 1H),$ 7.70 (m, 4H), 7.54 (d, J = 7.2 Hz, 2H), 1.33 (s, 9H); 13C NMR (151 MHz, cdcl₃) δ 165.12, 151.14, 148.19, 144.36, 138.68, 136.99, 136.34, 134.55, 133.41, 127.93, 127.74, 127.42, 127.15, 126.82, 125.87, 121.62, 121.58, 116.50, 34.57, 31.28; HRMS (ESI) m/z [M + H]⁺ calcd for C₂₆H₂₅N₂O 381.1967, found 381.1961.

4′-Methoxy-1-(quinolin-8-yl)[1,1′-biphenyl]-4-carboxamide (**5ad**): white solid (petroleum ether/acetone = 20:1, R_f = 0.15); 66.6 mg, yield 94%; mp 178−179 °C; ¹H NMR (600 MHz, DMSO) δ 10.72 (s, 1H), 9.01 (s, 1H), 8.77 (d, J = 7.2 Hz, 1H), 8.49 (d, J = 7.8 Hz, 1H), 8.10 $(d, J = 7.8 \text{ Hz}, 2H), 7.89 \ (d, J = 7.8 \text{ Hz}, 2H), 7.75 \ (d, J = 7.8 \text{ Hz}, 3H),$ 7.69 (d, J = 8.4 Hz, 2H), 7.09 (d, J = 7.8 Hz, 2H), 3.83 (s, 3H); ¹³C NMR $(150 \text{ MHz}, \text{CDCl}_3)$ δ 165.11, 159.69, 148.21, 144.11, 138.71, 136.31, 134.57, 133.02, 132.33, 128.22, 127.93, 127.75, 127.40, 126.78, 121.61, 121.56, 116.45, 114.33, 55.35; HRMS (ESI) m/z [M + H]⁺ calcd for $C_{23}H_{19}N_2O_2$ 355.1447, found 355.1441.

4′-Fluoro-1-(quinolin-8-yl)[1,1′-biphenyl]-4-carboxamide (5ae): white solid (petroleum ether/acetone = 20:1, R_f = 0.15); 65.1 mg, yield 95%; mp 152−153 °C; ¹ H NMR (600 MHz, DMSO) δ 10.73 (s, 1H), 9.01 (s, 1H), 8.77 (d, J = 7.2 Hz, 1H), 8.48 (d, J = 8.4 Hz, 1H), 8.13 $(d, J = 7.8 \text{ Hz}, 2H), 7.92 (d, J = 7.8 \text{ Hz}, 2H), 7.87–7.81 (m, 2H), 7.77 (d,$ $J = 8.4$ Hz, 1H), 7.74–7.65 (m, 2H), 7.37 (t, J = 8.4 Hz, 2H); ¹³C NMR $(150 \text{ MHz}, \text{DMSO}) \delta 164.04, 162.33 \text{ (d, } J = 244.2 \text{ Hz}), 149.17, 142.52,$ 138.23, 136.77, 135.37, 134.00, 133.10, 129.02 (d, J = 8.4 Hz), 127.83, 127.71, 127.12, 127.04, 122.34, 122.31, 116.52, 115.91 (d, J = 21.2 Hz); HRMS (ESI) m/z [M + H]⁺ calcd for C₂₂H₁₆FN₂O 343.1247, found 343.1261

3′-Fluoro-1-(quinolin-8-yl)[1,1′-biphenyl]-4-carboxamide (5af): white solid (petroleum ether/acetone = 20:1, R_f = 0.15); 63.0 mg, yield 92%; mp 93–94 °C; $^1\rm H\, NMR$ (600 MHz, DMSO) δ 10.74 (s, 1H), 9.02 (s, 1H), 8.76 (d, J = 7.2 Hz, 1H), 8.49 (d, J = 7.8 Hz, 1H), 8.14 (d, J $= 7.8$ Hz, 2H), 7.98 (d, J = 7.8 Hz, 2H), 7.77 (d, J = 7.8 Hz, 1H), 7.68 (m, 4H), 7.63−7.54 (m, 1H), 7.29 (t, J = 7.8 Hz, 1H); 13C NMR (150 MHz, CDCl₃) δ 164.84, 163.14 (d, J = 245.0 Hz), 148.21, 143.10, 142.15 (d, J $= 7.7$ Hz), 138.64, 136.38, 134.42, 134.26, 130.39 (d, J = 8.1 Hz), 127.93, 127.86, 127.40, 127.33, 122.79, 121.71, 121.64, 116.58, 114.78 (d, J = 20.9 Hz), 114.05 (d, J = 22.2 Hz); HRMS (ESI) m/z [M + H]⁺ calcd for $C_{22}H_{16}FN_2O$ 343.1247, found 343.1257.

2′-Fluoro-1-(quinolin-8-yl)[1,1′-biphenyl]-4-carboxamide (5ag): white solid (petroleum ether/acetone = 20:1, R_f = 0.15); 61.6 mg, yield 90%; mp 139−140 °C; ¹H NMR (600 MHz, DMSO) δ 10.75 (s, 1H), 9.01 (s, 1H), 8.77 (d, J = 7.2 Hz, 1H), 8.49 (d, J = 7.8 Hz, 1H), 8.16 $(d, J = 7.8 \text{ Hz}, 2H), 7.82 (d, J = 7.8 \text{ Hz}, 2H), 7.78 (d, J = 8.4 \text{ Hz}, 1H),$ 7.73−7.62 (m, 3H), 7.51 (d, J = 6.0 Hz, 1H), 7.44−7.34 (m, 2H); ¹³C NMR (150 MHz, CDCl₃) δ 164.97, 159.72 (d, J = 247.5 Hz), 148.23, 139.26, 138.68, 136.34, 134.50, 134.13, 130.59, 129.67 (d, J = 8.1 Hz), 129.33 (d, J = 2.7 Hz), 127.94, 127.39, 124.50 (d, J = 3.5 Hz), 121.67, 121.64, 116.54, 116.22 (d, J = 22.7 Hz); HRMS (ESI) m/z [M + H]⁺ calcd for $C_{22}H_{16}FN_2O$ 343.1247, found 343.1260.

4′-Chloro-1-(quinolin-8-yl)[1,1′-biphenyl]-4-carboxamide (5ah): white solid (petroleum ether/acetone = 20:1, R_f = 0.15); 66.7 mg, yield 93%; mp 153−154 °C; ¹ H NMR (600 MHz, DMSO) δ 10.73 (s, 1H), 9.01 (s, 1H), 8.76 (d, J = 7.2 Hz, 1H), 8.49 (d, J = 8.4 Hz, 1H), 8.14 $(d, J = 8.4 \text{ Hz}, 2H)$, 7.95 $(d, J = 8.4 \text{ Hz}, 2H)$, 7.83 $(d, J = 8.4 \text{ Hz}, 2H)$, 7.77 (d, J = 7.8 Hz, 1H), 7.69 (d, J = 7.2 Hz, 2H), 7.60 (d, J = 8.4 Hz, 2H); 13C NMR (150 MHz, DMSO) δ 164.02, 149.20, 142.21, 138.26, 137.70, 136.78, 133.98, 133.44, 133.21, 129.04, 128.71, 127.84, 127.78, 127.17, 127.05, 122.37, 116.60; HRMS (ESI) m/z [M + H]⁺ calcd for $C_{22}H_{16}CIN_{2}O$ 359.0951, found 359.0945.

4′-Hydroxy-1-(quinolin-8-yl)[1,1′-biphenyl]-4-carboxamide (5ai): white solid (petroleum ether/acetone = 20:1, R_f = 0.15); 59.2 mg, yield 87%; mp 218−219 °C; ¹ H NMR (600 MHz, DMSO) δ 10.71 (s, 1H), 9.75 (s, 1H), 9.01 (s, 1H), 8.77 (d, J = 7.2 Hz, 1H), 8.48 (d, J = 8.4 Hz, 1H), 8.08 (d, J = 7.8 Hz, 2H), 7.85 (d, J = 7.8 Hz, 2H), 7.76 (d, J = 7.8 Hz, 1H), 7.70 (m, 2H), 7.64 (m, 2H), 6.91 (d, J = 7.8 Hz, 2H); ¹³C NMR (150 MHz, DMSO) δ 164.2, 158.0, 149.2, 143.7, 138.2, 136.8, 134.1, 132.0, 129.6, 128.2, 127.9, 127.7, 127.1, 126.3, 122.4, 122.2, 116.5, 116.0; HRMS (ESI) m/z [M + H]⁺ calcd for C₂₂H₁₇N₂O₂ 341.1290, found 341.1281.

N-(Quinolin-8-yl)-4′-(trifluoromethoxy)[1,1′-biphenyl]-4-carboxamide (5aj): white solid (petroleum ether/acetone = 20:1, R_f = 0.15); 74.3 mg, yield 91%; mp 150−151 °C; ¹H NMR (600 MHz, ĎMSO) *δ* 10.74 (s, 1H), 9.01 (s, 1H), 8.76 (d, J = 7.2 Hz, 1H), 8.49 (d, J = 7.8 Hz, 1H), 8.15 (d, $J = 7.2$ Hz, 2H), 7.96 (d, $J = 7.2$ Hz, 2H), 7.92 (d, $J = 7.8$ Hz, 2H), 7.77 (d, J = 7.8 Hz, 1H), 7.74−7.65 (m, 2H), 7.53 (d, J = 7.8 Hz, 2H); 13C NMR (150 MHz, CDCl3) δ 164.85, 149.12, 148.25, 143.04, 138.70, 138.65, 136.37, 134.46, 134.13, 128.55, 127.96, 127.89, 127.42, 127.34, 121.73, 121.66, 121.29, 119.61, 116.56 (OCF₃ carbons are merging with other peaks); HRMS (ESI) m/z [M + H]⁺ calcd for $C_{23}H_{16}F_3N_2O_2$ 409.1164, found 409.1173.

N-(Quinolin-8-yl)-4′-(trifluoromethyl)[1,1′-biphenyl]-4-carboxamide (5ak): white solid (petroleum ether/acetone = 20:1, $R_f = 0.15$); 67.5 mg, yield 86%; mp 157−158 °C; ¹H NMR (600 MHz, ĎMSO) δ 10.74 (s, 1H), 9.01 (s, 1H), 8.76 (s, 1H), 8.48 (s, 1H), 8.17 (s, 2H), 8.01 (s, 4H), 7.88 (s, 2H), 7.77 (s, 1H), 7.70 (s, 2H); 13C NMR (150 MHz, CDCl3) δ 164.78, 148.26, 143.45, 142.94, 138.68, 136.44, 134.64, 134.42, 127.97, 127.58, 127.49, 127.44, 126.85, 125.82 (q, J = 3.0 Hz), 125.04, 124.14 (q, J = 270.8 Hz), 123.24, 121.80, 121.69, 116.63; HRMS (ESI) m/z [M + H]⁺ calcd for C₂₃H₁₆F₃N₂O 393.1215, found 393.1221.

4′-Acetyl-1-(quinolin-8-yl)[1,1′-biphenyl]-4-carboxamide (5al): yellow solid (petroleum ether/acetone = 20:1, R_f = 0.15); 66.0 mg, yield 90%; mp 159−160 °C; ¹H NMR (600 MHz, DMSO) δ 10.75 (s, 1H), 9.02 (s, 1H), 8.77 (d, J = 7.2 Hz, 1H), 8.49 (d, J = 7.8 Hz, 1H), 8.17 $(d, J = 7.8 \text{ Hz}, 2H), 8.10 (d, J = 7.8 \text{ Hz}, 2H), 8.02 (d, J = 7.8 \text{ Hz}, 2H),$ 7.96 (d, J = 7.8 Hz, 2H), 7.78 (d, J = 8.4 Hz, 1H), 7.74−7.67 (m, 2H),

2.65 (s, 3H); ¹³C NMR (150 MHz, CDCl₃) δ 197.53, 164.75, 148.26, 144.35, 143.05, 138.67, 136.36, 136.33, 134.57, 134.40, 128.94, 127.93, 127.89, 127.53, 127.39, 127.28, 121.75, 121.68, 116.52, 26.63; HRMS (ESI) m/z [M + H]⁺ calcd for C₂₄H₁₉N₂O₂ 367.1447, found 367.1445.

Methyl 4′-(quinolin-8-ylcarbamoyl)[1,1′-biphenyl]-4-carboxylate (5am): white solid (petroleum ether/acetone = 20:1, R_f = 0.15); 66.5 mg, yield 87%; mp 151−152 °C; ¹ H NMR (600 MHz, DMSO) δ 10.74 (s, 1H), 9.01 (s, 1H), 8.76 (s, 1H), 8.48 (s, 1H), 8.16 (s, 2H), 8.09 (s, 2H), 8.00 (s, 2H), 7.95 (s, 2H), 7.77 (s, 1H), 7.70 (s, 2H), 3.89 (s, 3H); $13C$ NMR (150 MHz, CDCl₃) δ 166.7, 164.7, 148.2, 144.2, 143.1, 138.6, 136.3, 134.5, 134.4, 130.1, 129.5, 127.9, 127.8, 127.5, 127.4, 127.1, 121.7, 121.6, 116.5, 52.1; HRMS (ESI) m/z [M + H]⁺ calcd for C₂₄H₁₉N₂O₃ 383.1396, found 383.1389.

2′,4′-Dichloro-1-(quinolin-8-yl)[1,1′-biphenyl]-4-carboxamide (**5an**): white solid (petroleum ether/acetone = $20:1$, $R_f = 0.15$); 70.0 mg, yield 89%; mp 189−190 °C; ¹H NMR (600 MHz, DMSO) δ 10.74 (s, 1H), 9.00 (s, 1H), 8.76 (d, J = 7.8 Hz, 1H), 8.49 (d, J = 7.8 Hz, 1H), 8.14 $(d, J = 7.8 \text{ Hz}, 2\text{H}), 7.82 \text{ (s, 1H)}, 7.78 \text{ (d, } J = 8.4 \text{ Hz}, 1\text{H}), 7.69 \text{ (m, 3H)},$ 7.64−7.52 (m, 2H); ¹³C NMR (150 MHz, CDCl₃) δ 165.0, 148.3, 141.7, 138.8, 138.0, 136.4, 134.6, 134.5, 134.3, 133.2, 131.9, 129.9, 129.8, 128.0, 127.5, 127.3, 127.2, 121.8, 121.7 116.6; HRMS (ESI) m/z [M + H]⁺ calcd for $C_{22}H_{15}Cl_2N_2O$ 393.0561, found 393.0555.

3′,4′-Dichloro-1-(quinolin-8-yl)[1,1′-biphenyl]-4-carboxamide (5ao): white solid (petroleum ether/acetone = 20:1, R_f = 0.15); 70.8 mg, yield 90%; mp 158−159 °C; ¹H NMR (600 MHz, DMSO) δ 10.73 (s, 1H), 9.01 (s, 1H), 8.75 (d, J = 7.2 Hz, 1H), 8.49 (d, J = 7.8 Hz, 1H), 8.14 $(d, J = 7.8 \text{ Hz}, 2H)$, 8.09 $(s, 1H)$, 8.00 $(d, J = 7.2 \text{ Hz}, 2H)$, 7.85–7.75 (m, 3H), 7.69 (d, J = 7.8 Hz, 2H); ¹³C NMR (150 MHz, CDCl₃) δ 164.7, 148.3, 141.9, 139.9, 138.7, 136.4, 134.5, 134.4, 133.0, 132.2, 130.8, 128.9, 128.0, 127.4, 127.2, 126.3, 121.8, 121.7, 116.6; HRMS (ESI) m/z [M + $[H]^+$ calcd for $C_{22}H_{15}Cl_2N_2O$ 393.0561, found 393.0558.

3′,5′-Dichloro-1-(quinolin-8-yl)[1,1′-biphenyl]-4-carboxamide (5ap): white solid (petroleum ether/acetone = 20:1, R_f = 0.15); 69.2 mg, yield 88%; mp 158–159 °C; ¹H NMR (600 MHz, CDCl₃) (δ ppm): 10.73 (s, 1H), 9.01 (d, J = 3.6 Hz, 1H), 8.75 (d, J = 7.2 Hz, 1H), 8.49 (d, J $= 7.2$ Hz, 1H), 8.14 (m, 3H), 8.02 (d, J = 8.4 Hz, 2H), 7.87 (s, 1H), 7.78 $(d, J = 8.4 \text{ Hz}, 1H), 7.73-7.66 \text{ (m, 3H)}; {}^{13}C NMR (150 MHz, CDCl₃) \delta$ 164.68, 148.29, 142.93, 141.70, 138.69, 136.44, 135.47, 134.91, 134.38, 128.02, 127.85, 127.50, 127.46, 127.40, 125.70, 121.84, 121.71, 116.65; HRMS (ESI) m/z [M + H]⁺ calcd for C₂₂H₁₅Cl₂N₂O 393.0561, found 393.0553.

4-(Pyridin-4-yl)-1-(quinolin-8-yl)benzamide (5aq): white solid (petroleum ether/acetone = 20:1, R_f = 0.15); 57.3 mg, yield 88%; mp 201−202 °C; ¹ H NMR (600 MHz, DMSO) δ 10.76 (s, 1H), 9.01 (s, 1H), 8.76 (d, J = 6.0 Hz, 1H), 8.72 (s, 2H), 8.49 (d, J = 6.6 Hz, 1H), 8.19 $(d, J = 6.0 \text{ Hz}, 2H)$, 8.08 (s, 2H), 7.83 (s, 2H), 7.78 (d, J = 6.6 Hz, 1H), 7.70 (s, 2H); ¹³C NMR (150 MHz, CDCl₃) δ 164.5, 150.3, 148.3, 147.1, 141.3, 138.6, 136.4, 135.4, 134.3, 128.0, 127.9, 127.4, 127.3, 121.8, 121.7, 121.6, 116.5; HRMS (ESI) m/z [M + H]⁺ calcd for C₂₁H₁₆N₃O 326.1293, found 326.1289.

4-(Naphthalen-1-yl)-1-(quinolin-8-yl)benzamide(5ar): white solid (petroleum ether/acetone = 20:1, R_f = 0.15); 66.6 mg, yield 89%; mp 151−152 °C; ¹ H NMR (600 MHz, DMSO) δ 10.76 (s, 1H), 9.03 (s, 1H), 8.78 (d, J = 7.2 Hz, 1H), 8.49 (d, J = 7.8 Hz, 1H), 8.38 (s, 1H), 8.19 $(d, J = 7.8 \text{ Hz}, 2H)$, 8.08 (m, 4H), 7.98 (m, 2H), 7.78 (d, J = 7.8 Hz, 1H), 7.74−7.67 (m, 2H), 7.59 (d, J = 6.0 Hz, 2H); 13C NMR (150 MHz, CDCl3) δ 165.05, 148.20, 144.39, 138.66, 137.19, 136.38, 134.51, 133.71, 133.51, 132.86, 128.62, 128.28, 127.94, 127.84, 127.62, 127.58, 127.43, 126.45, 126.31, 126.17, 125.16, 121.64, 116.55; HRMS (ESI) m/z [M + H]⁺ calcd for C₂₆H₁₉N₂O 375.1497, found 375.1487.

4-(Dibenzo[b,d]furan-4-yl)-1-(quinolin-8-yl)benzamide (5as): white solid (petroleum ether/acetone = 20:1, R_f = 0.15); 70.5 mg, yield 85%; mp 189−190 °C; ¹H NMR (600 MHz, DMSO) δ 10.80 (s, 1H), 9.03 (s, 1H), 8.80 (d, J = 6.6 Hz, 1H), 8.50 (d, J = 7.8 Hz, 1H), 8.25 $(m, 4H)$, 8.20 (s, 2H), 7.84 (d, J = 6.6 Hz, 1H), 7.79 (t, J = 8.4 Hz, 2H), 7.72 (s, 2H), 7.60 (m, 3H), 7.47 (s, 1H); 13C NMR (150 MHz, CDCl3) δ 165.1, 156.2, 153.3, 148.3, 139.9, 138.8, 136.4, 134.6, 134.2, 129.1, 128.0, 127.7, 127.5, 127.4, 126.8, 125.1, 124.7, 124.0, 123.3, 122.9, 121.7, 120.7, 120.4, 116.7, 111.9; HRMS (ESI) m/z [M + H]⁺ calcd for $C_{28}H_{19}N_2O_2$ 415.1447, found 415.1435.

N-(Quinolin-8-yl)[1,1′-biphenyl]-3-carboxamide (5ba): white solid (petroleum ether/acetone = 20:1, R_f = 0.15); 61.6 mg, yield 95%; mp $82-83 \text{ °C}$;^{38 1}H NMR (600 MHz, DMSO) δ 10.74 (s, 1H), 8.99 (d, J = 3.0 Hz, 1H), 8.74 (d, J = 7.8 Hz, 1H), 8.48 (d, J = 7.8 Hz, 1H), 8.29 (s, 1H), 8.03 [\(d,](#page-10-0) J = 7.2 Hz, 1H), 7.97 (d, J = 7.2 Hz, 1H), 7.79 (t, J = 8.4 Hz, 3H), 7.73 (t, J = 7.8 Hz, 1H), 7.71−7.67 (m, 2H), 7.54 (t, J = 7.8 Hz, 2H), 7.45 (t, J = 7.2 Hz, 1H); ¹³C NMR (150 MHz, DMSO) δ 164.5, 149.2, 140.9, 139.3, 138.4, 136.7, 135.3, 134.0, 130.3, 129.6, 129.1, 128.0, 127.8, 127.0, 126.9, 125.9, 125.5, 122.5, 122.3, 117.0; HRMS (ESI) m/z $[M + H]^{+}$ calcd for $C_{22}H_{17}N_2O$ 325.1341, found 325.1336.

4′-tert-Butyl)-1-(quinolin-8-yl)[1,1′-biphenyl]-3-carboxamide (**5bb**): white solid (petroleum ether/acetone = 20:1, R_f = 0.15); 72.3 mg, yield 95%; mp 116−117 °C; ¹ H NMR (600 MHz, DMSO) δ 10.73 (s, 1H), 8.99 (d, $J = 3.0$ Hz, 1H), 8.75 (d, $J = 7.8$ Hz, 1H), 8.48 (d, $J = 8.4$ Hz, 1H), 8.27 (s, 1H), 8.01 (d, J = 7.8 Hz, 1H), 7.94 (d, J = 7.8 Hz, 1H), 7.78 (d, J = 8.4 Hz, 1H), 7.70 (m, 5H), 7.55 (d, J = 7.8 Hz, 2H), 1.34 (s, 9H); ¹³C NMR (150 MHz, DMSO) δ 164.5, 150.4, 149.1, 140.8, 138.3, 136.7, 136.4, 135.2, 134.0, 130.1, 129.6, 127.8, 127.0, 126.5, 125.8, 125.6, 125.2, 122.4, 122.3, 116.8, 34.23, 31.02; HRMS (ESI) m/z [M + H]⁺ calcd for $C_{26}H_{25}N_2O$ 381.1967, found 381.1960.

4′-Methoxy-1-(quinolin-8-yl)[1,1′-biphenyl]-3-carboxamide (**5bc**): white solid (petroleum ether/acetone = 20:1, R_f = 0.15); 66.6 mg, yield 94%; mp 74−75 °C; $^1\rm H\, NMR$ (600 MHz, DMSO) δ 10.72 (s, 1H), 8.99 (s, 1H), 8.74 (d, J = 7.2 Hz, 1H), 8.48 (d, J = 7.2 Hz, 1H), 8.24 (s, 1H), 7.97 (d, J = 6.0 Hz, 1H), 7.92 (d, J = 6.6 Hz, 1H), 7.82−7.64 (m, 6H), 7.10 (d, J = 7.2 Hz, 2H), 3.83 (s, 3H); 13C NMR (150 MHz, DMSO) δ 164.6, 159.3, 149.2, 140.6, 138.4, 136.7, 135.2, 134.1, 131.6, 129.8, 129.6, 128.1, 127.9, 127.0, 125.1, 124.9, 122.4, 122.3, 116.9, 114.5, 55.2; HRMS (ESI) m/z [M + H]⁺ calcd for C₂₃H₁₉N₂O₂ 355.1447, found 355.1440.

4′-Fluoro-1-(quinolin-8-yl)[1,1′-biphenyl]-3-carboxamide (5bd): white solid (petroleum ether/acetone = 20:1, $R_f = 0.15$); 63.7 mg, yield 93%; mp 141−142 °C; ¹ H NMR (600 MHz, DMSO) δ 10.73 (s, 1H), 8.99 (d, J = 4.2 Hz, 1H), 8.72 (d, J = 7.8 Hz, 1H), 8.48 (d, J = 8.4 Hz, 1H), 8.27 (s, 1H), 8.03 (d, J = 7.8 Hz, 1H), 7.95 (d, J = 7.8 Hz, 1H), 7.85 (dd, J = 8.4, 5.4 Hz, 2H), 7.78 (d, J = 8.4 Hz, 1H), 7.74−7.67 (m, 3H), 7.37 (t, J = 9.0 Hz, 2H); ¹³C NMR (150 MHz, DMSO) δ 164.53, 162.17 (d, J = 243.5 Hz), 149.21, 139.85, 138.47, 136.73, 135.81, 135.32, 134.04, 130.27, 129.65, 129.01 (d, J = 8.0 Hz),, 127.86, 126.99, 125.87, 125.46, 122.53, 122.31, 117.08, 115.90 (d, J = 21.1 Hz); HRMS (ESI) m/z [M + H]⁺ calcd for C₂₂H₁₆FN₂O 343.1247, found 343.1251.

4′-Chloro-1-(quinolin-8-yl)[1,1′-biphenyl]-3-carboxamide (5be): white solid (petroleum ether/acetone = 20:1, R_f = 0.15); 66.7 mg, yield 93%; mp 156−157 °C; ¹ H NMR (600 MHz, DMSO) δ 10.73 (s, 1H), 8.99 (d, J = 4.2 Hz, 1H), 8.72 (d, J = 7.8 Hz, 1H), 8.48 (d, J = 8.4 Hz, 1H), 8.30 (s, 1H), 8.05 (d, J = 7.8 Hz, 1H), 7.97 (d, J = 7.8 Hz, 1H), 7.84 (d, J = 8.4 Hz, 2H), 7.78 (d, J = 8.4 Hz, 1H), 7.73 (t, J = 7.8 Hz, 1H), 7.71−7.67 (m, 2H), 7.60 (d, J = 8.4 Hz, 2H); ¹³C NMR (150 MHz, CDCl3) δ 165.16, 148.28, 140.57, 138.64, 138.56, 136.34, 135.77, 134.36, 133.87, 130.20, 129.25, 129.01, 128.40, 127.90, 127.36, 126.09, 125.92, 121.79, 121.66, 116.55; HRMS (ESI) m/z [M + H]⁺ calcd for $C_{22}H_{16}CIN_2O$ 359.0951, found 359.0944.

Methyl 3′-(quinolin-8-ylcarbamoyl)[1,1′-biphenyl]-4-carboxylate (**5bf**): white solid (petroleum ether/acetone = 20:1, R_f = 0.15); 65.0 mg, yield 85%; mp 166−167 °C; ¹H NMR (600 MHz, DMSO) δ 10.74 (s, 1H), 8.99 (d, J = 3.6 Hz, 1H), 8.72 (d, J = 7.8 Hz, 1H), 8.48 (d, J = 8.4 Hz, 1H), 8.36 (s, 1H), 8.10 (t, J = 9.0 Hz, 3H), 8.04 (d, J = 7.8 Hz, 1H), 7.97 (d, J = 7.8 Hz, 2H), 7.77 (dd, J = 15.6, 7.8 Hz, 2H), 7.69 (t, J = 6.6) Hz, 2H), 3.90 (s, 3H); ¹³C NMR (150 MHz, CDCl₃) δ 166.79, 165.08, 148.27, 144.49, 140.60, 138.60, 136.40, 135.81, 134.32, 130.49, 130.16, 129.31, 129.27, 127.91, 127.38, 127.11, 126.45, 126.42, 121.82, 121.67, 116.62, 52.15. HRMS (ESI) m/z [M + H]⁺ calcd for C₂₄H₁₉N₂O₃ 383.1396, found 383.1395.

N-(Quinolin-8-yl)-4′-(trifluoromethyl)[1,1′-biphenyl]-3-carboxa*mide* (**5bg**): white solid (petroleum ether/acetone = 20:1, R_f = 0.15); 66.7 mg, yield 85%; mp 151−152 °C; ¹H NMR (600 MHz, DMSO) δ 10.74 (s, 1H), 8.98 (s, 1H), 8.72 (d, J = 7.5 Hz, 1H), 8.48 (d, J = 8.3 Hz, 1H), 8.36 (s, 1H), 8.10 (d, J = 7.4 Hz, 1H), 8.04 (d, J = 7.9 Hz, 3H), 7.89 $(d, J = 7.8 \text{ Hz}, 2\text{H}), 7.78 \text{ (t, } J = 8.0 \text{ Hz}, 2\text{H}), 7.69 \text{ (t, } J = 7.0 \text{ Hz}, 2\text{H}); {^{13}\text{C}}$ NMR (150 MHz, DMSO) δ 164.53, 162.98, 161.35, 149.21, 139.84,

138.46, 136.72, 135.79, 135.32, 134.03, 130.26, 129.64, 129.02, 128.97, 127.85, 126.98, 125.85, 125.44, 122.52, 122.29, 117.08, 115.96, 115.82 (CF₃ carbons are merging with other peaks); HRMS (ESI) m/z [M + $[H]^+$ calcd for $C_{23}H_{16}F_3N_2O$ 393.1215, found 393.1223.

2′,4′-Dichloro-1-(quinolin-8-yl)[1,1′-biphenyl]-3-carboxamide (5bh): white solid (petroleum ether/acetone = 20:1, R_f = 0.15); 69.2 mg, yield 88%; mp 139−140 °C; ¹H NMR (600 MHz, DMSO) δ 10.72 (s, 1H), 8.98 (d, J = 3.6 Hz, 1H), 8.72 (d, J = 7.8 Hz, 1H), 8.48 (d, J = 8.4 Hz, 1H), 8.17 (s, 1H), 8.08 (d, J = 7.8 Hz, 1H), 7.83 (d, J = 7.8 Hz, 1H), 7.75 (m, 3H), 7.68 (t, J = 7.8 Hz, 2H), 7.46 (t, J = 9.6 Hz, 1H), 7.28 (t, J = 8.4 Hz, 1H); 13C NMR (150 MHz, DMSO) δ 164.27, 149.19, 138.40, 136.73, 134.99, 134.91, 133.98, 132.34, 132.11, 132.08, 132.05, 132.02, 129.39, 127.84, 127.56, 126.99, 126.36, 123.97, 122.52, 122.32, 116.97, 112.33, 112.19, 104.82, 104.65, 104.48; HRMS (ESI) m/z [M + H]⁺ calcd for $C_{22}H_{15}Cl_2N_2O$ 393.0561, found 393.0553.

3-(Pyridin-4-yl)-1-(quinolin-8-yl)benzamide (5bi): white solid (petroleum ether/acetone = 20:1, R_f = 0.15); 56.6 mg, yield 87%; mp 151−152 °C; ¹ H NMR (600 MHz, DMSO) δ 10.74 (s, 1H), 8.99 (d, J = 3.0 Hz, 1H), 8.72 (d, J = 6.0 Hz, 2H), 8.48 (d, J = 7.2 Hz, 1H), 8.42 (s, 1H), 8.13 (d, J = 7.8 Hz, 1H), 8.10 (d, J = 7.8 Hz, 1H), 7.85 (d, J = 6.0 Hz, 2H), 7.81−7.77 (m, 2H), 7.72−7.66 (m, 2H), 7.66−7.60 (m, 1H); 13C NMR (150 MHz, CDCl3) ^δ 164.7, 150.2, 148.3, 147.3, 138.7, 138.6, 136.3, 136.0, 134.2, 132.0, 130.1, 129.5, 127.8, 127.3, 127.2, 126.2, 121.9, 121.6, 116.5; HRMS (ESI) m/z [M + H]⁺ calcd for C₂₁H₁₆N₃O 326.1293, found 326.1295.

4-Phenyl-1-(quinolin-8-yl)-1-naphthamide (5ca): white solid (petroleum ether/acetone = 20:1, R_f = 0.15); 70.4 mg, yield 94%; mp 72− 73 °C; ¹H NMR (600 MHz, CDCl₃) δ 10.48 (s, 1H), 9.08 (d, J = 7.2 Hz, 1H), 8.77 (s, 1H), 8.59 (d, J = 8.4 Hz, 1H), 8.21 (d, J = 7.8 Hz, 1H), 7.96 $(d, J = 7.8 \text{ Hz}, 2H), 7.67 \text{ (t, } J = 7.8 \text{ Hz}, 1H), 7.62-7.56 \text{ (m, } 2H), 7.54-$ 7.44 (m, 8H); ¹³C NMR (150 MHz, CDCl₃) δ 167.8, 148.3, 143.3, 140.2, 138.5, 136.4, 134.8, 134.1, 132.1, 130.7, 129.9, 128.3, 128.0, 127.6, 127.4, 127.1, 126.5, 125.9, 125.8, 124.9, 121.9, 121.7, 116.7; HRMS (ESI) m/z [M + H]⁺ calcd for C₂₆H₁₉N₂O 375.1497, found 375.1486.

4-(4-Methoxyphenyl)-1-(quinolin-8-yl)-1-naphthamide (5cb): white solid (petroleum ether/acetone = 20:1, R_f = 0.15); 76.0 mg, yield 94%; mp 136−137 °C; ¹H NMR (600 MHz, CDCl₃) δ 10.47 (s, 1H), 9.08 (d, J = 7.2 Hz, 1H), 8.76 (s, 1H), 8.59 (d, J = 9.0 Hz, 1H), 8.20 $(d, J = 7.8 \text{ Hz}, 1\text{H}), 8.00 (d, J = 8.4 \text{ Hz}, 1\text{H}), 7.95 (d, J = 7.2 \text{ Hz}, 1\text{H}),$ 7.68−7.64 (m, 1H), 7.62−7.56 (m, 2H), 7.50 (t, J = 6.6 Hz, 2H), 7.46 $(d, J = 8.4 \text{ Hz}, 3\text{H})$, 3.92 (s, 3H); ¹³C NMR (150 MHz, CDCl₃) δ 167.8, 159.2, 148.2, 143.0, 138.5, 136.3, 134.8, 133.8, 132.5, 132.3, 131.0, 130.7, 127.9, 127.4, 127.0, 126.6, 126.4, 125.9, 125.8, 125.0, 121.9, 121.7, 116.7, 113.8, 55.3; HRMS (ESI) m/z [M + H]⁺ calcd for C₂₇H₂₁N₂O₂ 405.1603, found 405.1597.

4-(4-Fluorophenyl)-1-(quinolin-8-yl)-1-naphthamide (5cc): white solid (petroleum ether/acetone = 20:1, R_f = 0.15); 73.0 mg, yield 93%; mp 143−144 °C; ¹H NMR (600 MHz, CĎCl₃) δ 10.48 (s, 1H), 9.08 (d, $J = 7.2$ Hz, 1H), 8.77 (s, 1H), 8.59 (d, $J = 8.4$ Hz, 1H), 8.21 (d, $J = 8.4$ Hz, 1H), 7.96 (d, J = 7.2 Hz, 1H), 7.91 (d, J = 8.4 Hz, 1H), 7.67 (t, J = 7.8 Hz, 1H), 7.63−7.57 (m, 2H), 7.53−7.46 (m, 5H), 7.22 (t, ^J = 7.8 Hz, 2H); 13C NMR (150 MHz, CDCl3) ^δ 167.7, 162.4 (d, ^J = 245.7 Hz), 148.3, 142.1, 138.5, 136.4, 136.1, 134.7, 134.3, 132.1, 131.5 (d, $J = 8.0$ Hz), 130.6, 128.0, 127.4, 127.1, 126.7, 126.2, 126.0, 125.8, 124.8, 122.0, 121.7, 116.8, 115.3 (d, J = 21.2 Hz); HRMS (ESI) m/z [M + H]⁺ calcd for $C_{26}H_{18}FN_{2}O$ 393.1403, found 393.1397.

4-(Pyridin-4-yl)-1-(quinolin-8-yl)-1-naphthamide (5cd): white solid (petroleum ether/acetone = 20:1, $R_f = 0.15$); 63.1 mg, yield 84%; mp 193−194 °C; ¹H NMR (600 MHz, CDCl₃) δ 10.48 (s, 1H), 9.07 (d, J = 7.2 Hz, 1H), 8.79 (s, 2H), 8.77 (s, 1H), 8.60 (d, J = 8.4 Hz, 1H), 8.22 (d, J = 8.4 Hz, 1H), 7.98 (d, J = 7.2 Hz, 1H), 7.88 (d, J = 8.4 Hz, 1H), 7.67 (t, J = 7.8 Hz, 1H), 7.63 (m, 2H), 7.56 (t, J = 7.8 Hz, 1H), 7.54−7.50 (m, 3H), 7.48 (dd, J = 7.8, 3.6 Hz, 1H); ¹³C NMR (150 MHz, CDCl3) δ 167.33, 149.76, 148.30, 148.22, 140.10, 138.50, 136.38, 135.32, 134.59, 131.24, 130.60, 127.96, 127.45, 127.39, 127.16, 126.05, 125.83, 125.64, 124.86, 124.70, 122.10, 121.72, 116.75; HRMS (ESI) m/z [M + H]⁺ calcd for C₂₅H₁₈N₃O 376.1450, found 376.1445.

5-Phenyl-N-(quinolin-8-yl)furan-2-carboxamide (6aa): white solid (petroleum ether/acetone = 20:1, $R_f = 0.15$); 57.8 mg, yield 92%; mp 159−160 °C; ¹ H NMR (600 MHz, CDCl3) δ 10.92 (s, 1H), 8.95 (d, J =

3.0 Hz, 1H), 8.89 (d, J = 7.2 Hz, 1H), 8.22 (d, J = 8.4 Hz, 1H), 7.88 (d, J $= 7.8$ Hz, 2H), 7.60 (d, J = 7.8 Hz, 1H), 7.57 (d, J = 8.4 Hz, 1H), 7.54− 7.47 (m, 3H), 7.39 (t, J = 5.4 Hz, 2H), 6.84 (d, J = 3.6 Hz, 1H); ¹³C NMR (150 MHz, CDCl₃) δ 156.23, 155.70, 148.39, 147.19, 138.52, 136.20, 134.14, 129.56, 128.82, 128.65, 127.89, 127.28, 124.49, 121.64, 121.59, 117.05, 116.48, 107.47; HRMS (ESI) m/z [M + H]⁺ calcd for $C_{20}H_{15}N_2O_2$ 315.1134, found 315.1140.

5-(4-Fluorophenyl)-N-(quinolin-8-yl)furan-2-carboxamide (6ab): white solid (petroleum ether/acetone =20:2, $R_f = 0.15$); 60.5 mg, yield 92%; mp 157−158 °C; ¹H NMR (600 MHz, CDCl₃) δ 10.88 (s, 1H), 8.94 (d, J = 3.6 Hz, 1H), 8.89 (d, J = 7.2 Hz, 1H), 8.23 (d, J = 8.4 Hz, 1H), 7.90−7.82 (m, 2H), 7.61 (t, J = 7.8 Hz, 1H), 7.57 (d, J = 7.8 Hz, 1H), 7.52 (dd, J = 8.4, 4.2 Hz, 1H), 7.39 (d, J = 3.6 Hz, 1H), 7.19 (t, J = 8.4 Hz, 2H), 6.78 (d, J = 3.6 Hz, 1H); ¹³C NMR (150 MHz, CDCl₃) δ 162.8 (d, J = 248.0 Hz), 156.1, 154.8, 148.4, 147.2, 138.5, 136.2, 134.1, 127.9, 127.3, 126.40 (d, $J = 8.3$ Hz), 126.0, 121.7, 121.6, 117.1, 116.5, 116.0 (d, $J = 21.0$ Hz), 107.2; HRMS (ESI) m/z [M + H]⁺ calcd for $C_{20}H_{14}FN_{2}O_{2}$ 333.1039, found 333.1040.

5-(Pyridin-4-yl)-N-(quinolin-8-yl)furan-2-carboxamide (6ac): white solid (petroleum ether/acetone =20:3, $R_f = 0.15$); 52.4 mg, yield 83%; mp 187−188 °C; ¹ H NMR (600 MHz, DMSO) δ 10.96 (s, 1H), 8.96 (d, J = 3.0 Hz, 1H), 8.88 (d, J = 7.2 Hz, 1H), 8.75 (d, J = 5.4 Hz, 2H), 8.23 (d, J = 8.4 Hz, 1H), 7.75 (d, J = 5.4 Hz, 2H), 7.60 (q, J = 8.4 Hz, 2H), 7.54 (dd, J = 8.4, 4.2 Hz, 1H), 7.42 (d, J = 3.6 Hz, 1H), 7.08 (d, J = 3.6 Hz, 1H); ¹³C NMR (150 MHz, CDCl₃) δ 155.61, 152.59, 150.34, 148.60, 148.49, 138.44, 136.23, 136.21, 133.79, 127.86, 127.22, 121.95, 121.69, 118.18, 116.76, 116.56, 110.75; HRMS (ESI) m/z [M + $[H]^+$ calcd for $C_{19}H_{14}N_3O_2$ 316.1086, found 316.1089.

5-Phenyl-N-(quinolin-8-yl)thiophene-2-carboxamide (6ad): white solid (petroleum ether/acetone = 20:1, R_f = 0.15); 50.2 mg, yield 76%; mp 135−136 °C; ¹H NMR (600 MHz, DMSO) δ 10.59 (s, 1H), 9.01 (s, 1H), 8.60 (d, J = 7.8 Hz, 1H), 8.48 (d, J = 7.8 Hz, 1H), 8.03 (s, 1H), 7.81 $(d, J = 7.8 \text{ Hz}, 2H), 7.77 \, (d, J = 8.4 \text{ Hz}, 1H), 7.70 \, (s, 2H), 7.67 \, (t, J = 7.8 \,$ Hz, 1H), 7.49 (t, J = 7.8 Hz, 2H), 7.42 (t, J = 7.2 Hz, 1H); ¹³C NMR $(150 \text{ MHz}, \text{CDCl}_3)$ δ 159.82, 149.77, 148.24, 138.36, 136.32, 134.19, 133.38, 129.38, 129.00, 128.55, 127.88, 127.36, 126.07, 123.59, 121.66, 121.59, 116.39; HRMS (ESI) m/z [M + H]⁺ calcd for C₂₀H₁₅N₂OS 331.0905, found 331.0910.

5-(4-Fluorophenyl)-N-(quinolin-8-yl)thiophene-2-carboxamide (6ae): white solid (petroleum ether/acetone =20:2, R_f = 0.15); 53.0 mg, yield 76%; mp 169−170 °C; ¹H NMR (600 MHz, DMSO) δ 10.58 (s, 1H), 9.00 (s, 1H), 8.59 (d, J = 7.8 Hz, 1H), 8.48 (d, J = 7.8 Hz, 1H), 8.02 $(s, 1H)$, 7.85 (d, J = 6.0 Hz, 2H), 7.77 (d, J = 8.4 Hz, 1H), 7.71–7.62 (m, 3H), 7.34 (t, J = 8.4 Hz, 2H); ¹³C NMR (150 MHz, CDCl₃) δ 162.80 (d, J = 247.7 Hz), 159.66, 148.54, 148.22, 138.43, 138.34, 136.28, 134.14, 129.69, 129.67, 129.33, 127.86, 127.79 (d, J = 8.1 Hz), 127.32, 123.54, 121.62 (d, J = 5.3 Hz), 116.37, 115.97 (d, J = 21.9 Hz); HRMS (ESI) $m/$ z [M + H]⁺ calcd for C₂₀H₁₄FN₂OS 349.0811, found 349.0817.

5-(Pyridin-4-yl)-N-(quinolin-8-yl)thiophene-2-carboxamide (6af): white solid (petroleum ether/acetone = 20:3, R_f = 0.15); 47.1 mg, yield 71%; mp 194−195 °C; ¹ H NMR (600 MHz, DMSO) δ 10.64 (s, 1H), 9.01 (s, 1H), 8.66 (s, 2H), 8.59 (d, J = 7.2 Hz, 1H), 8.48 (d, J = 8.4 Hz, 1H), 8.10 (s, 1H), 7.96 (s, 1H), 7.79 (s, 3H), 7.70 (s, 1H), 7.57 (s, 1H); ¹³C NMR (150 MHz, CDCl₃) δ 159.3, 150.5, 148.4, 146.0, 141.0, 140.5, 138.4, 136.4, 134.0, 129.2, 127.9, 127.4, 125.7, 121.9, 121.8, 120.0, 116.6; HRMS (ESI) m/z [M + H]⁺ calcd for C₁₉H₁₄N₃OS 332.0858, found 332.0852.

5-Phenyl-N-(quinolin-8-yl)thiazole-2-carboxamide (6ag): white solid (petroleum ether/acetone = 20:3, R_f = 0.15); 49.7 mg, yield 75%; mp 157−158 °C; ¹ H NMR (600 MHz, CDCl3) δ 10.92 (s, 1H), 8.95 (d, J = 3.6 Hz, 1H), 8.90 (d, J = 7.8 Hz, 1H), 8.21 (d, J = 7.8 Hz, 1H), 7.88 (d, J = 7.8 Hz, 2H), 7.60 (t, J = 7.8 Hz, 1H), 7.56 (d, J = 8.4 Hz, 1H), 7.51 (m, 3H), 7.40 (t, J = 6.0 Hz, 2H), 6.84 (d, J = 3.6 Hz, 1H); ¹³C NMR (150 MHz, CDCl₃) δ 156.3, 155.7, 148.4, 147.2, 138.5, 136.3, 134.1, 129.6, 128.8, 128.7, 127.9, 127.3, 124.5, 121.7, 121.6, 117.1, 116.6, 107.5; HRMS (ESI) m/z [M + H]⁺ calcd for C₁₉H₁₄N₃OS 332.0858, found 332.0861.

5-(4-Fluorophenyl)-N-(quinolin-8-yl)thiazole-2-carboxamide (6ah): white solid (petroleum ether/acetone = 20:4, R_f = 0.15); 51.0 mg, yield 73%; mp 157−158 °C; ¹H NMR (600 MHz, CDCl₃) δ 10.88 (s,

1H), 8.93 (d, $J = 3.0$ Hz, 1H), 8.89 (d, $J = 7.2$ Hz, 1H), 8.21 (d, $J = 7.8$ Hz, 1H), 7.85 (dd, J = 8.4, 5.4 Hz, 2H), 7.60 (t, J = 7.8 Hz, 1H), 7.56 (d, J $= 7.8$ Hz, 1H), 7.51 (dd, J = 8.4, 4.2 Hz, 1H), 7.37 (d, J = 3.6 Hz, 1H), 7.19 (t, J = 8.4 Hz, 2H), 6.77 (d, J = 3.6 Hz, 1H); ¹³C NMR (150 MHz, CDCl₃) δ 162.8 (d, J = 247.8 Hz), 156.1, 154.8, 148.3, 147.2, 138.4, 136.3, 134.0, 127.9, 127.3, 126.4 (d, J = 8.3 Hz), 125.9, 121.7, 121.6, 117.1, 116.6, 116.0 (d, J = 21.9 Hz), 107.2; HRMS (ESI) m/z [M + H]⁺ calcd for $C_{19}H_{13}FN_3OS$ 350.0763, found 350.0780.

5-(Pyridin-4-yl)-N-(quinolin-8-yl)thiazole-2-carboxamide (6ai): white solid (petroleum ether/acetone =20:4, $R_f = 0.15$); 46.5 mg, yield 70%; mp 186−187 °C; ¹H NMR (600 MHz, CDCl₃) δ 10.96 (s, 1H), 8.96 (s, 1H), 8.88 (d, J = 7.2 Hz, 1H), 8.75 (d, J = 4.8 Hz, 2H), 8.23 $(d, J = 8.4 \text{ Hz}, 1H)$, 7.75 $(d, J = 4.8 \text{ Hz}, 2H)$, 7.64–7.56 $(m, 2H)$, 7.54 $(dd, J = 8.4, 4.2 \text{ Hz}, 1\text{H}), 7.42 \text{ (d, } J = 3.6 \text{ Hz}, 1\text{H}), 7.08 \text{ (d, } J = 3.6 \text{ Hz},$ 1H); ¹³C NMR (150 MHz, CDCl₃) δ 155.6, 152.6, 150.3, 148.6, 148.5, 138.5, 136.3, 133.8, 127.9, 127.2, 122.0, 121.7, 118.2, 116.8, 116.6, 110.8; HRMS (ESI) m/z [M + H]⁺ calcd for C₁₈H₁₃N₄OS 333.0810, found 333.0813.

N-(Pyridin-2-ylmethyl)[1,1′-biphenyl]-4-carboxamide (7aa): white solid (petroleum ether/acetone = 20:1, R_f = 0.15); 54.2 mg, yield 94%; mp 124–125 °C;³⁹ ¹H NMR (600 MHz, DMSO) δ 9.20 (s, 1H), 8.52 $(d, J = 4.8 \text{ Hz}, 1H), 8.03 (d, J = 7.8 \text{ Hz}, 2H), 7.81 (d, J = 7.8 \text{ Hz}, 2H),$ 7.76 (m, 3H), 7.5[0 \(](#page-10-0)t, J = 7.2 Hz, 2H), 7.43 (d, J = 7.2 Hz, 1H), 7.34 (d, J $= 7.8$ Hz, 1H), 7.31–7.24 (m, 1H), 4.60 (d, J = 6.0 Hz, 2H); ¹³C NMR (150 MHz, DMSO) δ 166.0, 158.8, 148.9, 142.9, 139.2, 136.7, 133.0, 129.0, 128.1, 128.0, 126.9, 126.6, 122.1, 120.9, 44.8; HRMS (ESI) m/z $[M + H]^{+}$ calcd for $C_{19}H_{17}N_{2}O$ 289.1341, found 289.1365.

4′-Fluoro-N-(pyridin-2-ylmethyl)[1,1′-biphenyl]-4-carboxamide (**7ab**): white solid (petroleum ether/acetone = 20:1, R_f = 0.15); 57.0 mg, yield 93%; mp 171−172 °C; ¹ H NMR (600 MHz, DMSO) δ 9.20 (t, J = 5.4 Hz, 1H), 8.52 (d, J = 4.2 Hz, 1H), 8.02 (d, J = 7.8 Hz, 2H), 7.84–7.73 $(m, 5H)$, 7.33 (t, J = 7.2 Hz, 3H), 7.30–7.26 (m, 1H), 4.60 (d, J = 6.0 Hz, 2H).¹³C NMR (150 MHz, DMSO) δ 166.0, 162.3 (d, J = 243.6 Hz), 158.9, 148.9, 141.8, 136.7, 135.6, 132.9, 129.0 (d, J = 8.1 Hz), 128.0, 126.6, 122.1, 120.9, 115.9 (d, J = 21.3 Hz), 44.8; HRMS (ESI) m/z [M + H]⁺ calcd for C₁₉H₁₆FN₂O 307.1247, found 307.1239.

4′-Chloro-N-(pyridin-2-ylmethyl)[1,1′-biphenyl]-4-carboxamide (7ac): white solid (petroleum ether/acetone = 20:1, R_f = 0.15); 60.1 mg, yield 93%; mp 193−194 °C; ¹ H NMR (600 MHz, DMSO) δ 9.21 (s, 1H), 8.52 (s, 1H), 8.02 (d, J = 6.6 Hz, 2H), 7.84–7.73 (m, 5H), 7.55 (d, J $= 3.6$ Hz, 2H), 7.33 (d, J = 8.4 Hz, 1H), 7.28 (s, 1H), 4.59 (d, J = 5.4 Hz, 2H); 13C NMR (150 MHz, DMSO) δ 165.9, 158.8, 148.9, 141.5, 138.0, 136.8, 133.3, 133.0, 129.0, 128.7, 128.1, 126.6, 122.1, 120.9, 44.8; HRMS (ESI) m/z $[M + H]^{+}$ calcd for $C_{19}H_{16}CIN_{2}O$ 323.0951, found 323.0942.

4′-Methoxy-N-(pyridin-2-ylmethyl)[1,1′-biphenyl]-4-carboxamide (**7ad**): white solid (petroleum ether/acetone = 20:1, R_f = 0.15); 59.9 mg, yield 94%; mp 139−140 °C; ¹H NMR (600 MHz, DMSO) *δ* 9.18 (t, J = 6.0 Hz, 1H), 8.52 (d, J = 4.2 Hz, 1H), 7.99 (d, J = 8.4 Hz, 2H), 7.77 (m, 3H), 7.71 (d, J = 9.0 Hz, 2H), 7.33 (d, J = 7.8 Hz, 1H), 7.30−7.25 (m, 1H), 7.06 (d, J = 9.0 Hz, 2H), 4.59 (d, J = 6.0 Hz, 2H), 3.81 (s, 3H); ¹³C NMR (150 MHz, DMSO) δ 166.1, 159.4, 158.9, 148.9, 142.6, 136.8, 132.2, 131.4, 128.1, 128.0, 126.0, 122.1, 120.9, 114.5, 55.2, 44.8; HRMS (ESI) m/z [M + H]⁺ calcd for C₂₀H₁₉N₂O₂ 319.1447, found 319.1438.

4′-tert-Butyl-N-(pyridin-2-ylmethyl)[1,1′-biphenyl]-4-carboxa*mide (7ae):* white solid (petroleum ether/acetone = 20:1, $R_f = 0.15$); 65.4 mg, yield 95%; mp 102−103 °C; ¹H NMR (600 MHz, DMSO) δ 9.18 (t, J = 6.0 Hz, 1H), 8.52 (d, J = 4.8 Hz, 1H), 8.01 (d, J = 8.4 Hz, 2H), 7.80−7.73 (m, 3H), 7.67 (d, J = 8.4 Hz, 2H), 7.51 (d, J = 8.4 Hz, 2H), 7.34 (d, J = 7.8 Hz, 1H), 7.31−7.23 (m, 1H), 4.60 (d, J = 6.0 Hz, 2H), 1.32 (s, 9H); 13C NMR (150 MHz, DMSO) δ 166.1, 158.9, 150.6, 148.9, 142.8, 136.7, 136.3, 132.7, 128.0, 126.6, 126.4, 125.8, 122.1, 120.9, 44.7, 34.3, 31.1; HRMS (ESI) m/z [M + H]⁺ calcd for C₂₃H₂₅N₂O 345.1967, found 345.1958.

N-(Pyridin-2-ylmethyl)-4′-(trifluoromethyl)[1,1′-biphenyl]-4-carboxamide (7af): white solid (petroleum ether/acetone = 20:1, R_f = 0.15); 61.3 mg, yield 86%; mp 187−188 °C; ¹ H NMR (600 MHz, DMSO) δ 9.27 (t, J = 6.0 Hz, 1H), 8.53 (d, J = 4.8 Hz, 1H), 8.07 (d, J = 8.4 Hz, 2H), 7.98 (d, J = 8.4 Hz, 2H), 7.88 (m, 4H), 7.77 (t, J = 7.2 Hz, 1H), 7.34 (d, J = 7.8 Hz, 1H), 7.30−7.25 (m, 1H), 4.61 (d, J = 6.0 Hz, 2H); 13C NMR (150 MHz, DMSO) δ 165.9, 158.8, 148.9, 143.2, 141.2,

136.7, 133.9, 131.4, 129.5, 128.2, 127.7, 127.1, 125.9, 125.2, 123.4, 122.1, 120.9, 44.8 ($CF₃$ carbons are merging with other peaks); HRMS (ESI) m/z [M + H]⁺ calcd for C₂₀H₁₆F₃N₂O 357.1215, found 357.1207.

2′,4′-Dichloro-N-(pyridin-2-ylmethyl)[1,1′-biphenyl]-4-carboxamide (**7ag**): white solid (petroleum ether/acetone = 20:1, $R_f = 0.15$); 62.9 mg, yield 88%; mp 106−107 °C; ¹H NMR (600 MHz, DMSO) δ 9.26 (s, 1H), 8.52 (d, J = 4.1 Hz, 1H), 8.02 (d, J = 8.0 Hz, 2H), $7.84 - 7.68$ $(m, 2H)$, 7.56 (d, J = 8.0 Hz, 3H), 7.49 (d, J = 8.3 Hz, 1H), 7.35 (d, J = 7.6 Hz, 1H), 7.31−7.25 (m, 1H), 4.60 (d, J = 5.7 Hz, 2H); 13C NMR $(150 \text{ MHz}, \text{CDCl}_3)$ δ 165.93, 158.73, 148.86, 140.40, 137.99, 136.71, 133.71, 133.34, 132.65, 132.27, 129.33, 129.26, 127.74, 127.31, 122.08, 120.87, 44.73; HRMS (ESI) m/z [M + H]⁺ calcd for C₁₉H₁₅Cl₂N₂O 357.0561, found 357.0551.

PYR1 Computational Modeling. Molecular docking studies were performed to simulate the binding of compound to PYR1. Conformational optimizations were performed on A, which was used as the starting structure for docking. The PYR1 crystal structure (PDB: 3QN1) was prepared as follows: (1) water and ligand were removed; (2) polar hydrogen atoms were added; (3) a grid box for the binding site was created (center $x = 0.943$, center $y = 22.758$, center $z = 33.926/\text{size } x =$ 18, size $y = 18$, size $z = 18$). Docking calculations were performed on it with AutoDock4.0.2. The protein and ligand structures were prepared with AutoDock Tools.3. The atomic Gasteiger−Huckel charges were assigned to the ligand and receptor. A total of 256 runs were launched for each compound. Most of the parameters for the docking calculation were set to the default values recommended by the software. Each docked structure was scored by the built-in scoring function and was clustered according to RMSD < 2 Å.

■ ASSOCIATED CONTENT

S Supporting Information

The Supporting Information is available free of charge on the ACS Publications website at DOI: 10.1021/acs.joc.6b01725.

X-ray data of compound 5ba (CIF) ¹

 1 H and 13 C NMR spectra an[d X-ray crystallographic d](http://pubs.acs.org/doi/abs/10.1021/acs.joc.6b01725)ata (PDF)

■ A[UTHO](http://pubs.acs.org/doi/suppl/10.1021/acs.joc.6b01725/suppl_file/jo6b01725_si_002.pdf)R INFORMATION

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Notes

The aut[hors declare no competing](mailto:gfyang@mail.ccnu.edu.cn) financial interest.

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■ REFERENCES

(1) Senger, J.; Melesina, J.; Marek, M.; Romier, C.; Oehme, I.; Witt, O.; Sippl, W.; Jung, M. J. Med. Chem. 2016, 59, 1545−1555.

(2) Zhao, H.; Garg, G.; Zhao, J.; Moroni, E.; Girgis, A.; Franco, L. S.; Singh, S.; Colombo, G.; Blagg, B. S. J. Eur. J. Med. Chem. 2015, 89, 442− 466.

(3) Ohashi, M.; Oyama, T.; Putranto, E. W.; Waku, T.; Nobusada, H.; Kataoka, K.; Matsuno, K.; Yashiro, M.; Morikawa, K.; Huh, N. H.; Miyachi, H. Bioorg. Med. Chem. 2013, 21, 2319−2332.

(4) Ravu, V. R.; Leung, G. Y. C.; Lim, C. S.; Ng, S. Y.; Sum, R. J.; Chen, D. Y. K. Eur. J. Org. Chem. 2011, 2011, 463-468.

(5) Liang, J.; van Abbema, A.; Balazs, M.; Barrett, K.; Berezhkovsky, L.; Blair, W.; Chang, C.; Delarosa, D.; DeVoss, J.; Driscoll, J.; Eigenbrot, C.; Ghilardi, N.; Gibbons, P.; Halladay, J.; Johnson, A.; Kohli, P. B.; Lai, Y.; Liu, Y.; Lyssikatos, J.; Mantik, P.; Menghrajani, K.; Murray, J.; Peng, I.;

Sambrone, A.; Shia, S.; Shin, Y.; Smith, J.; Sohn, S.; Tsui, V.; Ultsch, M.; Wu, L. C.; Xiao, Y.; Yang, W.; Young, J.; Zhang, B.; Zhu, B. Y.; Magnuson, S. J. Med. Chem. 2013, 56, 4521−4536.

(6) Shimizu, H.; Tanaka, S.; Toki, T.; Yasumatsu, I.; Akimoto, T.; Morishita, K.; Yamasaki, T.; Yasukochi, T.; Iimura, S. Bioorg. Med. Chem. Lett. 2010, 20, 5113−5118.

(7) Mor, M.; Rivara, S.; Lodola, A.; Plazzi, P. V.; Tarzia, G.; Duranti, A.; Tontini, A.; Piersanti, G.; Kathuria, S.; Piomelli, D. J. Med. Chem. 2004, 47, 4998−5008.

(8) Mach, U. R.; Lewin, N. E.; Blumberg, P. M.; Kozikowski, A. P. ChemMedChem 2006, 1, 307−314.

(9) Vachal, P.; Miao, S.; Pierce, J. M.; Guiadeen, D.; Colandrea, V. J.; Wyvratt, M. J.; Salowe, S. P.; Sonatore, L. M.; Milligan, J. A.; Hajdu, R.; Gollapudi, A.; Keohane, C. A.; Lingham, R. B.; Mandala, S. M.; DeMartino, J. A.; Tong, X.; Wolff, M.; Steinhuebel, D.; Kieczykowski, G. R.; Fleitz, F. J.; Chapman, K.; Athanasopoulos, J.; Adam, G.; Akyuz, C. D.; Jena, D. K.; Lusen, J. W.; Meng, J.; Stein, B. D.; Xia, L.; Sherer, E. C.; Hale, J. J. J. Med. Chem. 2012, 55, 2945−2959.

(10) Kasai, S.; Kamata, M.; Masada, S.; Kunitomo, J.; Kamaura, M.; Okawa, T.; Takami, K.; Ogino, H.; Nakano, Y.; Ashina, S.; Watanabe, K.; Kaisho, T.; Imai, Y. N.; Ryu, S.; Nakayama, M.; Nagisa, Y.; Takekawa, S.; Kato, K.; Murata, T.; Suzuki, N.; Ishihara, Y. J. Med. Chem. 2012, 55, 4336−4351.

(11) Lanier, M.; Schade, D.; Willems, E.; Tsuda, M.; Spiering, S.; Kalisiak, J.; Mercola, M.; Cashman, J. R. J. Med. Chem. 2012, 55, 697− 708.

(12) Williams, J. D.; Khan, A. R.; Cardinale, S. C.; Butler, M. M.; Bowlin, T. L.; Peet, N. P. Bioorg. Med. Chem. 2014, 22, 419−434.

(13) Westaway, S. M.; Thompson, M.; Rami, H. K.; Stemp, G.; Trouw, L. S.; Mitchell, D. J.; Seal, J. T.; Medhurst, S. J.; Lappin, S. C.; Biggs, J.; Wright, J.; Arpino, S.; Jerman, J. C.; Cryan, J. E.; Holland, V.; Winborn, K. Y.; Coleman, T.; Stevens, A. J.; Davis, J. B.; Gunthorpe, M. J. Bioorg. Med. Chem. Lett. 2008, 18, 5609−5613.

(14) Westaway, S. M.; Chung, Y. K.; Davis, J. B.; Holland, V.; Jerman, J. C.; Medhurst, S. J.; Rami, H. K.; Stemp, G.; Stevens, A. J.; Thompson, M.; Winborn, K. Y.; Wright, J. Bioorg. Med. Chem. Lett. 2006, 16, 4533− 4536.

(15) For selected recent examples of condensation between acid and 8 aminoquinoline, see: (a) Yan, Q.; Chen, Z.; Yu, W.; Yin, H.; Liu, Z.; Zhang, Y. Org. Lett. 2015, 17, 2482−2485. (b) Shibata, K.; Chatani, N. Chem. Sci. 2016, 7, 240−245. (c) Aihara, Y.; Chatani, N. J. Am. Chem. Soc. 2013, 135, 5308−5311.

(16) Singh, B. K.; Jana, R. J. Org. Chem. 2016, 81, 831−841.

(17) Huang, Z. Y.; Yang, J. F.; Chen, Q.; Cao, R. J.; Huang, W.; Hao, G. F.; Yang, G. F. RSC Adv. 2015, 5, 75182−75186.

(18) (a) Tietze, L. F.; Beifuss, U. Angew. Chem., Int. Ed. Engl. 1993, 32, 131. (b) Tietze, L. F. Chem. Rev. 1996, 96, 115. (c) Broadwater, S. J.; Roth, S. L.; Price, K. E.; Kobašlija, M.; McQuade, D. T. Org. Biomol. Chem. 2005, 3, 2899. (d) Sydnes, M. Curr. Green Chem. 2014, 1, 216.

(19) For selected recent examples of palladium-catalyzed reactions to biaryl carboxamides, see: (a) Miyaura, N.; Suzuki, A. Chem. Rev. 1995, 95, 2457−2483. (b) Stanforth, S. P. Tetrahedron 1998, 54, 263−303. (c) Alberico, D.; Scott, M. E.; Lautens, M. Chem. Rev. 2007, 107, 174− 238. (d) Ackermann, L.; Vicente, R.; Kapdi, A. R. Angew. Chem., Int. Ed. 2009, 48, 9792−9826. (e) Aihara, Y.; Chatani, N. Chem. Sci. 2013, 4, 664−670. (f) Stibingerova, I.; Voltrova, S.; Kocova, S.; Lindale, M.; Srogl, J. Org. Lett. 2016, 18, 312−315. (g) Kleeb, S.; Pang, L.; Mayer, K.; Eris, D.; Sigl, A.; Preston, R. C.; Zihlmann, P.; Sharpe, T.; Jakob, R. P.; Abgottspon, D.; et al. J. Med. Chem. 2015, 58, 2221−2239. (h) Barrett, K. T.; Miller, S. Org. Lett. 2015, 17, 580−583.

(20) For selected recent examples of the N-8-quinolyl group chelating the catalyst, see: (a) Zaitsev, V. G.; Shabashov, D.; Daugulis, O. J. Am. Chem. Soc. 2005, 127, 13154−13155. (b) Aihara, Y.; Chatani, N. Chem. Sci. 2013, 4, 664−670. (c) Shibata, K.; Chatani, N. Org. Lett. 2014, 16, 5148−5151. (d) Yokota, A.; Aihara, Y.; Chatani, N. J. Org. Chem. 2014, 79, 11922−11932.

(21) (a) Liu, Y. C.; Huang, Z. Y.; Chen, Q.; Yang, G. F. Tetrahedron 2013, 69, 9025−9032. (b) Qu, R. Y.; Liu, Y. C.; Wu, Q. Y.; Chen, Q.; Yang, G. F. Tetrahedron 2015, 71, 8123−8130. (c) Liu, Y. C.; Ye, C. J.; Chen, Q.; Yang, G. F. Tetrahedron Lett. 2013, 54, 949−955. (d) Liu, Y. C.; Qu, R. Y.; Chen, Q.; Wu, Q. Y.; Yang, G. F. Tetrahedron 2014, 70, 2746−2752. (e) Zhou, Z.; Zhao, P.; Huang, W.; Yang, G. Adv. Synth. Catal. 2006, 348, 63−67. (f) Zhou, Z. Z.; Yang, G. F. Bioorg. Med. Chem. 2006, 14, 8666−8674. (g) Zhou, Z. Z.; Ji, F. Q.; Cao, M.; Yang, G. F. Adv. Synth. Catal. 2006, 348, 1826−1830.

(22) For selected recent examples of the deprotection of amide N-Boc, see: (a) El Kazzouli, S.; Koubachi, J.; Berteina-Raboin, S.; Mouaddib, A.; Guillaumet, G. Tetrahedron Lett. 2006, 47, 8575. (b) Wang, J.; Liang, Y. L.; Qu, J. Chem. Commun. 2009, 5144.

(23) Hao, G. F.; Zhu, X. L.; Ji, F. Q.; Zhang, L.; Yang, G. F.; Zhan, C. G. J. Phys. Chem. B 2009, 113, 4865−4875.

(24) Raha, K.; Merz, K. M. J. Med. Chem. 2005, 48, 4558.

(25) Pan, Y.; Gao, D.; Zhan, C. G. J. Am. Chem. Soc. 2008, 130, 5140.

(26) Grigorjeva, L.; Daugulis, O. Org. Lett. 2014, 16, 4684−4687.

(27) Katayev, D.; Pfister, K. F.; Wendling, T.; Gooßen, L. J. Chem. - Eur. J. 2014, 20, 9902−9905.

(28) Zhang, S. Y.; Li, Q.; He, G.; Nack, W. A.; Chen, G. J. Am. Chem. Soc. 2015, 137, 531−539.

(29) Takamatsu, K.; Hirano, K.; Miura, M. Org. Lett. 2015, 17, 4066− 4069.

(30) Talbot, E. P.; Fernandes, T. d. A.; McKenna, J. M.; Toste, F. D. J. Am. Chem. Soc. 2014, 136, 4101−4104.

(31) Nishino, M.; Hirano, K.; Satoh, T.; Miura, M. Angew. Chem. 2013, 125, 4553−4557.

(32) Tran, L. D.; Popov, I.; Daugulis, O. J. Am. Chem. Soc. 2012, 134, 18237.

(33) Ano, Y.; Tobisu, M.; Chatani, N. Org. Lett. 2012, 14, 354.

(34) Goldfarb, S. D. (The University of Rochester) Patent US20090163545, 2016; p 18.

(35) Dahl, R. (Eiger biopharmaceutical, Inc.) Patent WO2016032569 A1, 2016; p 132.

(36) Inoue, S.; Shiota, H.; Fukumoto, Y.; Chatani, N. J. Am. Chem. Soc. 2009, 131, 6898.

(37) Gui, Q.; Chen, X.; Hu, L.; Wang, D.; Liu, J.; Tan, Z. Adv. Synth. Catal. 2016, 358, 509.

(38) Kubo, T.; Chatani, N. Org. Lett. 2016, 18, 1698.

(39) Shiota, H.; Ano, Y.; Aihara, Y.; Fukumoto, Y.; Chatani, N. J. Am. Chem. Soc. 2011, 133, 14952.